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LiCu₂O₂,

01.04.07

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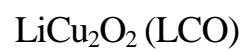
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2.7.			57
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3.5.3.2		$\text{Li}(\text{Cu}_{1-x}\text{Zn}_x)_2\text{O}_2$	105
			107
			110



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 LiCu_2O_2
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 LiCu_2O_2 .
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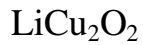
$$\rho_{DC} = A \cdot \exp(T_0/T)^{1/4}.$$

~25 DC

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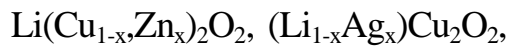
3 .



a, b ***c***: *a* : *b* :

$$c = 2 : 1 : 10^4.$$

3 .



$$(x(\text{Zn}) \approx 0,05, x(\text{Ag}) < 0,02)$$

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LiCu_2O_2 $\text{Li}(\text{Cu,Zn})_2\text{O}_2$,
 $(\text{Li,Ag})\text{Cu}_2\text{O}_2$:

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LiCu_2O_2

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 CuO_4^-

(ladder compound),

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 LiCu_2O_2

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 LiCu_2O_2

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1. . . DC AC
 $\text{LiCu}_2\text{O}_{2+}$ / . . , . . , . . [.] //
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 .174–178.
 2. .
 , LiCu_2O_2 /
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 . . // . 2015. .51. – 6. – .660–668 (Hieu
 Sy Dau. Effect of silver solubility on the structural, electrical, and magnetic
 properties of multiferroic LiCu_2O_2 / Hieu Sy Dau, K.E. Kamentsev, V.P.
 Sirotinkin, K.A. Yakovlev, E.A. Tishchenko, A.A. Bush // Inorganic Materials,
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 3. . .
 LiCu_2O_2 / . . , . . , . .
 , . . , . . // . 2015.
 – 5. – . 716–720 (Sirotinkin V. P. X-Ray Diffraction Analysis of LiCu_2O_2 crystals
 with additives of silver atoms / V.P. Sirotinkin, A.A. Bush, K.E. Kamentsev, H.S. Dau,
 K.A. Yakovlev, and E.A. Tishchenko // Crystallography Reports. 2015. – Vol. 60. –
 5. – P. 662–666).
 4. . .
 LiCu_2O_2
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 LiCu_2O_2

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6. . . DC AC
 $\text{LiCu}_2\text{O}_{2+}$ / . . , , . . , . .
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7. . .
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9. . . , ,
 $\text{Li}(\text{Cu}_{1-x}\text{Ag}_x)_2\text{O}_2$ / . . , . . , . . , . .
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1.

1.1

1.1.1

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[1–3].

[4, 5],

[6–8],

[9–13],

[14]

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 [15, 16].
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 $\text{K}_{0.3}\text{MoO}_3$
 [17].
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 [15].
 S– ,
 [18, 19].
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 [20]. ,
 . ,

[15, 19, 20].

3d-

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CuO_4 ,

SrCuO_2 , Sr_2CuO_3 ,

SrCu_2O_3 $\text{Sr}_{14}\text{Cu}_{24}\text{O}_{41}$ [21–24].

[25].

[26, 27],

[24].

(. .) [28].

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：， S-
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· · [22, 24, 29].

1.1.2

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[30, 31].
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[30].
1933
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(
 r_p)
 a .
， $r_p >> a$.

$, r_p < a,$
 $r_p \sim a.$

[9–13],
(S=1/2),
(n p)

[33].

(PA – phonon assisted). C
PA

(), kT ,
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[34 – 42].
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() [30].
, , (N —
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[31].

1.1.3

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« » (,),
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(. .) ,
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 μ
(DOS).
, . . .
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 R .
« » ,
 $\ln R \sim T^l$.

$lnR \sim T^n, \quad n < 1$ [43–45].

[43 – 45].

$p_1(1-p_2)+p_2(1-p_1), \quad p_1 \quad p_2 -$

$(\quad \quad \quad [46] \quad [47]).$

$(\quad \quad \quad) \quad \quad \quad :$
 $(\quad \quad \quad) \quad \quad \quad)$
 $(\quad \quad \quad).$
 $(\quad \quad \quad) \quad E_a$

$(\quad_{max} - \mu, \quad_{max} + \mu) [44].$

[44]

, -

.

i)

, :

$$\dagger = \dagger \exp\left(-\frac{v}{k_B T}\right), \tag{1.1}$$

v .

ii)

,

(. 1.1).

p

:

) $\exp(-/k_B T)$, -

, k_B- .

) , ,

) , (

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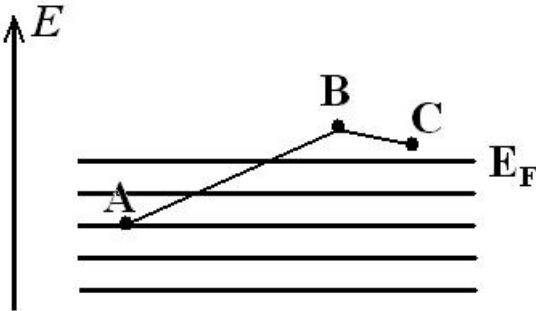
$\exp(-2 R)$, $1/$ - ,

,

,

1).

:



1.1.

, A () B B C [44].

$$\dagger \approx \left(\frac{e^2}{k_B T} \right) p R^2 N(E_F) k_B T \approx e^2 p R^2 N(E_F), \quad (1.2)$$

$$p = \epsilon \exp \left(-2r R - \frac{v}{k_B T} \right), \quad (1.3)$$

$$R - \quad , N(E_F) - \quad . \quad N(E).$$

, :

$$v \approx \frac{1}{R^3 N(E)} \quad (1.4)$$

,

.

$$R + [44]:$$

$$\left(\frac{4f}{3} \right) R^3 N(v) dv, \quad (1.5)$$

R

$$\Delta v = \frac{3}{4f R^3 N(E_F)}, \quad (1.6)$$

$$\epsilon \exp \left\{ -2r R - \left[\left(\frac{4f}{3} \right) R^3 N(E_F) k_B T \right]^{-1} \right\}. \quad (1.7)$$

$$(1.7) \quad R_{opt}$$

$$2r = \left[\frac{9}{4} f R^4 N(E_F) k_B T \right]^{-1} \quad (1.8)$$

$$R_{opt} \quad (1.7) \quad (1.8),$$

$$\epsilon = \exp\left(-\frac{B}{T^{1/4}}\right), \quad B \approx 2,1 \left[\frac{\mathfrak{r}^3}{k_B N(E_F)} \right]^{1/4}, \tag{1.9}$$

$$(1.2) \quad (1.3) \quad ,$$

$$\mathfrak{t} = \mathfrak{t}_o \left(\frac{T_o}{T} \right)^{1/4}, \tag{1.10}$$

$$T_M \qquad [36, 48]$$
$$T = 2,1^4 . [{}^3/k_B N(E_F)] \tag{1.12}$$

:

$$N(E_F)$$

.

,

,

$$| \quad - \mu | \quad ,$$

$$n(\quad) = 2N(E_F) \quad .$$

$$(1.8) \quad n(\quad)$$

$$R_{opt} \quad [n(\quad)]^{-1/3} = [(9/2) \quad^{-1} N(E_F) k_B T]^{-1/4}, \tag{1.13}$$

$$_{opt}$$

$$(1.13)$$

$$_{opt} = [(9/2) \quad]^{3/4} ((k \quad)^{3/4} [N(E_F)/{}^3]^{1/4} = k_B \quad^{3/4} T_o^{1/4}, \tag{1.14}$$

$$, \qquad (1.13) \quad (1.14)$$

$$(\quad = \quad_o \exp\{E_{\alpha}/k_B T\}) \quad ,$$

$$(\quad) \quad , \qquad ,$$

$$_{max} (\quad)$$

$$_{min} (\quad) \quad .$$

$$_{max}/k_B = T_0^{1/4} . T^{-3/4} \tag{1.15}$$

$$T \qquad \qquad \qquad (\ln \ , \ T^{1/4}) \qquad (\ln \ , \ T^{-1}),$$

$$\dots \qquad \qquad \qquad \dots \qquad \qquad \qquad [43] \text{ (SE)} \qquad \qquad \qquad ,$$

$$\qquad \qquad \qquad - \qquad \qquad \qquad - \qquad \qquad \qquad .$$

$$max \qquad \qquad \qquad , \qquad \qquad \qquad , \qquad \qquad \qquad max \qquad \qquad \qquad ,$$

$$\dagger = \dagger_0 \exp \left(- \frac{T_{SE}}{T} \right)^{1/2}, \tag{1.16}$$

$$- \qquad \qquad \qquad T_{SE}$$

$$T_{SE} = \frac{s_{SE} e^2 r}{|k_B|}, \tag{1.17}$$

$$_{SE} = 2,8, \qquad - \qquad \qquad \qquad .$$

$$/k_B = (TT_{SE})^{1/2}.$$

$$T < T_V, \qquad \qquad \qquad_{opt} (T)$$

$$(T_V) = k_B(T_V T_{SE})^{1/2},$$

$$- \qquad \qquad \qquad T_V, \qquad \qquad \qquad ,$$

$$\qquad \qquad \qquad (\ , \),$$

$$(\ln \ , \ T^{-1/2}).$$

$$,$$

$$[37, 43, 44]$$

$$\dagger = \dagger_0 \exp \left(- \frac{T_0}{T} \right)^{\epsilon}, \tag{1.18}$$

$$\dagger_0 = AT^{-m}, \tag{1.19}$$

$$A \quad m - \qquad \qquad \qquad .$$

,
 ,
 $= 1 - 1/4$,
(1.5), (1.7) (1.8),

,
 , $= 1/(d+1)$, $d -$.
 m (1.19)

- [43] $3 -$
 $m_{opt} >$, [37] SE m_{opt} .
 $m = (m_{opt} a/2 - s)^2$,

-
 , $s = \sqrt{q} -$,
 . $m_{opt} a/2 - s \sim qa$ $qa > 1$

-
 ,
 ,
 .

$F(r) \sim \exp(-r/a)$ $= 1/2 - m(\ll 1) = 1/2$
 $m(\gg 1) = 9/2;$ $= 1/4 - m(\ll 1) = 1/4$ $m(\gg 1) = 25/4.$
 $F(r) \sim r^{-1} \exp(-$
 $r/a)$ $= 1/2 - m(\ll 1) = -3/2$ $m(\gg 1) = 5/2;$ $= 1/4 - m(\ll 1) =$
 $- 3/4$ $m(\gg 1) = 21/4$ [37].

(SP). ,
 ,
 -

W. ,
 [48]. . .

W.

$$w(r, W) = \exp\{-2r/a - (r + W/2)/T\}, \quad (1.20)$$

$$= -1 -$$

r

:

$$r = N^{1/d} = (G W)^{-1/d}, \quad (1.21)$$

$$N - , d = 2 \quad 3$$

G

(DOS) ,

$$G(W_o, W_o).d .dW -$$

$$(r, r + d) \quad W \quad (W_o, W_o + dW). \quad G \quad (1.21)$$

$$(1.20) \quad , \quad w(r, W) \quad W$$

:

$$r_{opt} = W_{opt}/2 = T[T_o^{(d)}/T]^{2/(d+2)}, \quad r_{opt} = (a/2)[T_o^{(d)}/T]^{2/(d+2)}, \quad (1.22)$$

$$T_o^{(d)} = [r_o^{(d)}/Ga^d]^{1/2},$$

$$(1.23)$$

$$r_o^{(d)} = 21,1 \quad 31,2 \quad d = 2 \quad 3, \quad ,$$

-

.

SP

$$(1.18) \quad (1.19), \quad = 2/(d+2) \quad :$$

$$= r_o \exp\{-[T_o^{(d)}/T]^{2/(d+2)}\} \quad (1.24)$$

(W_{min},

$$W_{max}), \quad > T_1^{(d)}$$

$$W_{opt}(r) > W_{max} = W_{opt}(r_1^{(d)}), \quad :$$

$$r_1^{(d)} = \{[t_o^{(d)}W_{max}/4]^{d+1}/T_M^{(d)}\}^{1/d}, \quad T_M^{(d)} = r_M^{(d)}/ga^d \quad (1.25)$$

$$t_o^{(d)} = 0,546 \quad 0,607 \quad r_M^{(d)} = 13,8 \quad 17,0$$

$$d = 2 \quad 3, \quad , \quad g - \quad \text{DOS.}$$

SP

()

:

$$= \exp\{-[T_M^{(d)}/T]^{1/(d+1)} - {}_dW_{max}/T\}, \quad (1.26)$$

$${}_d = 0,189 \quad 0,174 \quad d = 2$$

3.

(1.26),

,

.

 g , G

-

.

$$= = [(e^2/k)^d g]^{1/(d-1)} = [(e^2/ka)^d T_M^{(d)}]^{1/(d-1)}, \quad (1.27)$$

SE

$$< T_2^{(d)},$$

$$> {}_{opt}(T_2^{(d)})$$

$$T_2^{(d)} = [{}_o/T_o^{(d)}]^{2/d}, \quad (1.28)$$

,

$$[T_2^{(d)} < T < {}_I^{(d)}],$$

,

,

.

,

$$W_{min}, \quad (1.24)$$

$$, \quad W_{opt}(T_3^{(d)}) < W_{min}, \quad T_3^{(d)} = (W_{min}/W_{max})^{(d+1)/d} T_I^{(d)}.$$

(1.24)

$$\max[T_2^{(d)}, T_3^{(d)}] < T < {}_I^{(d)},$$

$$, \quad (W_{max} - W_{min})/{}_o \gg 1.$$

$$, \quad [2(T^d T_M^{(d)})]/W_{min} \ll 1 \quad \ll T_3^{(d)},$$

:

$$= {}_o \exp\{-W_{min}/2T - [s_d T_M^{(d)}/T]^{1/(d+1)}\}, \quad (1.29)$$

$$s_d = 0,643$$

$$d = 2$$

$$0,63$$

$$d = 3.$$

$$W_{max} - W_{min}$$

 opt ,

,

(1.24) $W_{max} - W_{min}$

$W_{max} >$ (1.26),

(1.29).

,

$m,$

(1.18) (1.19).

()

$E_D = -d \ln / d(1/k_B T)$ (1.18) (1.19) :

$\ln[(E_D/k_B T) + m] = \ln + \ln T_{oj} + \ln(1/T),$ (1.30)

,

m (1.30)

$\ln(1/T),$

$T_{oj} (, T_{SE} T_o^{(d)})$

.

m, T_{oj} ,

(1.18) (1.19),

.

$= 1 \quad m = 1 (\quad) \quad m = 3/2$

().

,

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[44].

LCMFO [37]

LSMFO [47, 50]) m

(),
[37]. ,

[37, 47], $m = 25/4$ $9/2$

- .
,

,
(DOS)

.
DOS .
,
. 1.2

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, _____

[45].

,
SP, DOS
 $g(E_F) \sim 0$, ,

. $W_{\min}/2$.
,

,
 $2E_d$,

$= W_{\min}/2 - E_d/2,$ (1.32)



1.2.

[43].

() < .

(. .)

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 1,
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,
 [50, 51].
 (d.c.),

, . .
 ,
 .

$$\begin{aligned}
 & (\quad) , \\
 & (\quad) \sim C^{-s} \tag{1.31}
 \end{aligned}$$

$s \approx 0,8$ [43, 44]. C

$$\begin{aligned}
 & [44] , \\
 & W_1 - W_2 \quad (W = W_1 - W_2),
 \end{aligned}$$

$$W \quad n$$

$$D, \quad F$$

$$\frac{nDF \cos^2 \theta}{k_B T (1 + \tilde{S}^2 \dagger^2)} \left\{ 1 + \exp \left(\frac{\Delta W}{k_B T} \right) \right\}^{-1} \tag{1.32}$$

$F = \cos^2$, $1/3$.

$$\dagger(\check{S}) = \frac{nD^2\check{S}^2\dagger}{3k_B T} (1 + \check{S}^2\dagger^2)^{-1} \left\{ 1 + \exp\left(\frac{\Delta W}{k_B T}\right) \right\}^{-1} \quad (1.33)$$

$$W = 0 \quad Nd(W) \quad W \quad d(W),$$

W :

$$\int N \left\{ 1 + \exp\left(\frac{\Delta W}{k_B T}\right) \right\}^{-1} d(\Delta W) \quad (1.34)$$

$$N \quad NkT \ln 2, \quad :$$

$$\dagger(\check{S}) = \frac{(\ln 2) ND^2\check{S}^2\dagger}{3(1 + \check{S}^2\dagger^2)} \quad (1.35)$$

[44],

U , :

$$\dagger(\check{S}) = \frac{f}{6} (\ln 2) NBk_B T \check{S} D^2 \quad (1.36)$$

$B =$.

, R ,

:

$$\dagger(\check{S}) = A \frac{e^2}{r^3} \{N(E_F)\}^2 k_B T \check{S} \left\{ \ln\left(\frac{\epsilon_{ph}}{\check{S}}\right) \right\}^4, \quad (1.37)$$

$$A = (\sqrt{2}/24) \ln 2 \approx 0.3.$$

$$(\) \sim s^s (s \approx 1)$$

,

.

$$, \quad ph \quad 10^{12} \text{ }^{-1},$$

$$[\ln(\epsilon_{ph}/\)]^4 \quad -0.2 \quad 10^4$$

, $s \approx 0.8 -$.

$s = 2$

- [44].
Bottger Bryksin [46],

[43] .

A.R. Long [51]

,

C - (1.19)

:

$(\) \sim T^{n-s}$ (1.38)

$s \quad n$,

$(\)$

$(\)$

tg

[35, 51 – 53].

,

$\ddagger(T_{max}) \approx 1$ (1.39)

[44, 52, 54],

$(= 1/2 \ f),$:

$=_{oi}\exp(E_a/kT)$ (1.40)

$f_o = 1/2 \ o$

, $E_a -$

.

tg

[52, 53] $(tg)_{\max} \sim n_0$.

1.2

LiCu₂O₂

1.2.1

LiCu₂O₂,

,

YBa₂Cu₃O_y [55–58].

[59]

LiCu₂O₂ [56]Cu⁺S=1/2 Cu²⁺

,

.

. Pnma (. . 62, Z = 4),

 $a = 5,7286(2)$, $b = 2,8588(1)$, $c = 12,4143(3)$ Å

[55].

 a/b

,

-

()

,

,

,

 ab –

(.

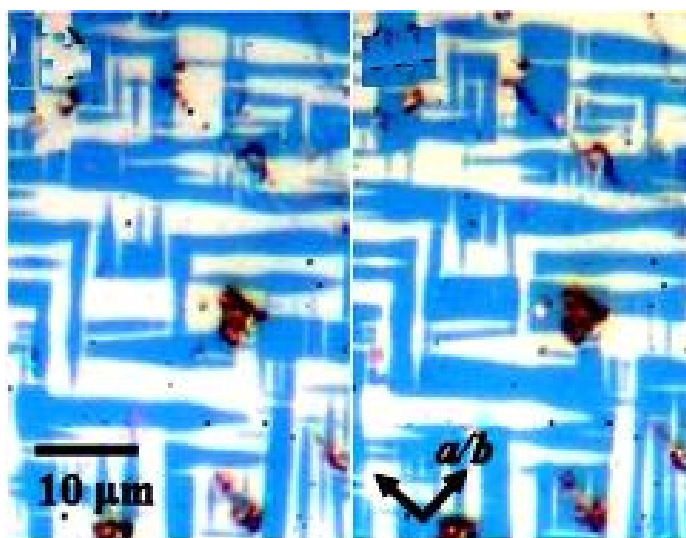
1.3).

. 1.4

LiCu₂O₂,

,

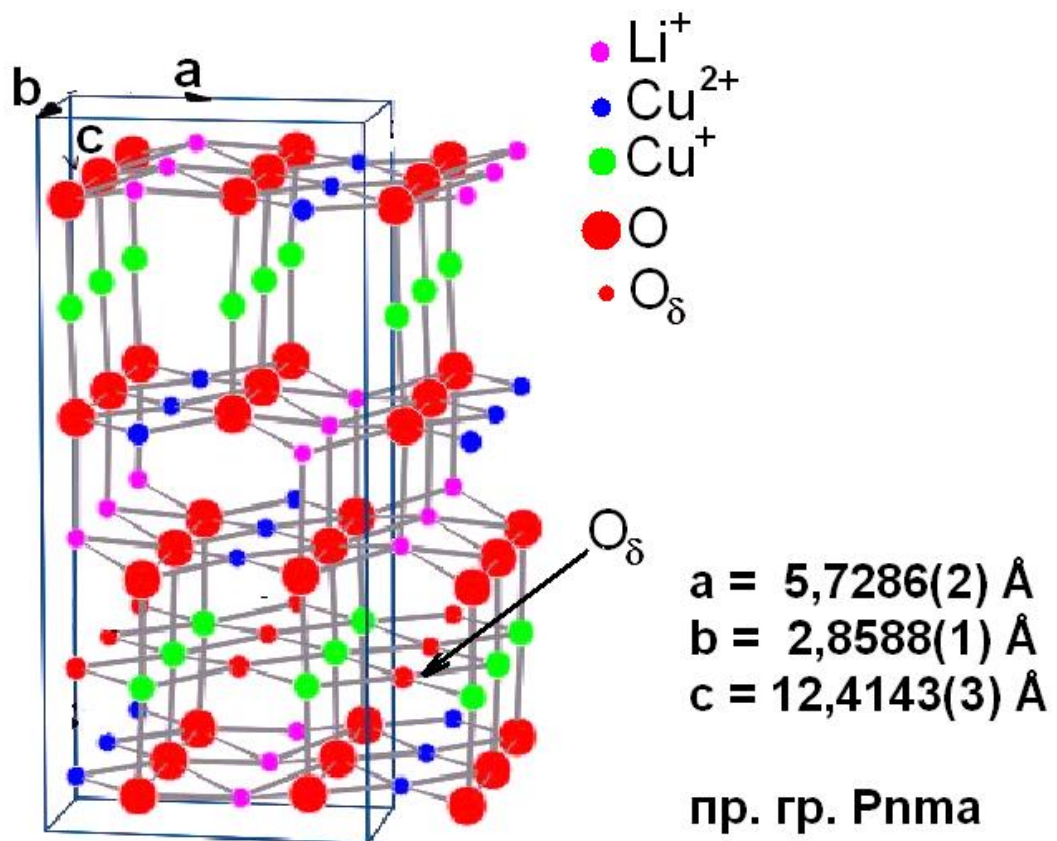
[55, 56, 58, 59].



1.3.

LiCu₂O₂,

[55].



1.4.

 LiCu_2O_2 LiCu_2O_2 : 1) $-\text{Cu}^{1+}(1)-$, 2) $-\text{O}(1)\text{Cu}^{2+}(2)\text{O}(2)\text{Li}-$ 3) - $\text{LiO}(2)\text{Cu}^{2+}(2)\text{O}(1).$ Cu^{1+} $\text{O}^{2-}-\text{Cu}^{1+}-\text{O}^{2-}$ LiCuO_2-

2) 3)

 CuO_5 LiO_5 ,*ab*- LiO_4 CuO_4 *a* Cu-O- Li- O- *b* - Cu-O- Li-O- Cu-O- CuO_5- Cu-O (1,98 Å),

(2,48 Å),

 Cu^{2+} [59]. LiO_5 Li-O

(~2,08 Å).

1) Cu^{1+} d^9

,
 O^{2-}
 O^{2-}
 $\text{C}^{n+} = \text{Cu}^{2+}$ Li^{1+}
 CO_5 CO_6
 LiCu_2O_2 ,
b
 (ladder system).
 Li^+
 Cu^+ (. 1.4)
 LiCu_2O_2
 $(S = 1/2)$
 [59-63].
 Cu^{2+} Li^+
 u-O-Cu
 $\text{O}2\text{p}$,
 $\text{O}2\text{p-Cu}3\text{d}$ CuO_4
 Pmna
 [56, 57, 64].

1.2.2

LiCu₂O₂



-



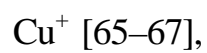
b.



S=1/2 (two-leg ladder systems).

-

,



.

NN (J_1)

NNN

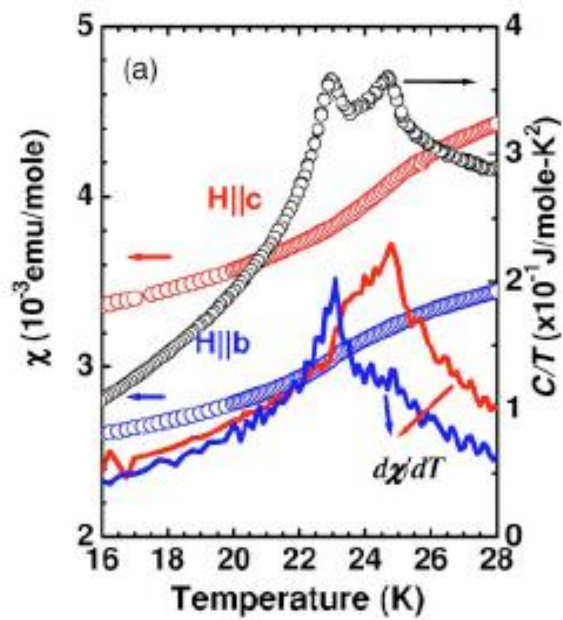
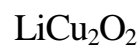
(J_2)

–

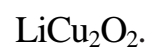
,

(J_3) ($J_1 = -7,0$, $J_2 = 3,75$, $J_3 = 3,4$

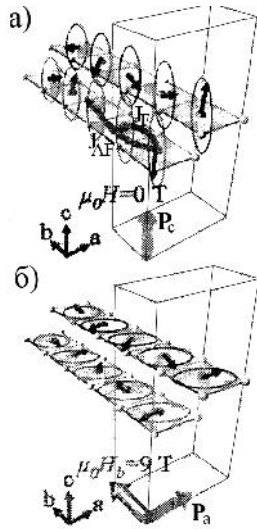
[66, 67]).



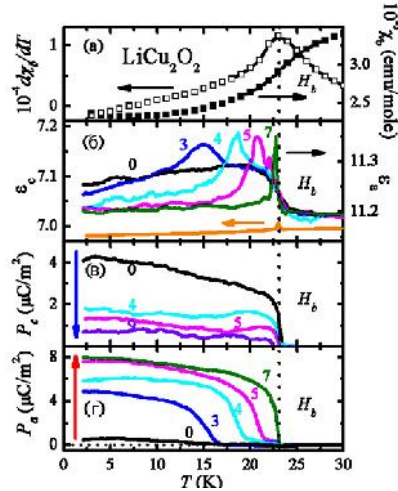
1.5.



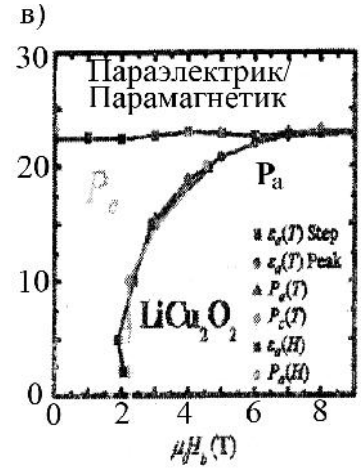
$T_{c1}=24,6$ $T_{c2}=23,2$ [68].



1.6.
1.6.



1.7.



1.8

Cu^{2+}

Cu-O
()

$H=9,0 \text{ T}$,
1.7. LiCu_2O_2 :

$b-$ () [65, 69]).
: (a)

; ()
()

b

$H_b=2$

c a

H_b ; ()

c

a

H_b (

) [65, 69].

1.8.

LiCu_2O_2 ,

P_c

P_a ,

H_b .

Cu-O

$T_{c1} = 24,6$

$T_{c2} = 23,2$ (1.5)

:

T_{c1} T_{c2}

T_{c2} [65–74].

[65, 68, 69, 72, 73, 75–78]

,

LiCu_2O_2

.

c

P_s .

T_{c2}

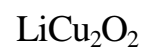
-

P_s

($\sim 1.6 - 1.8$).



II ().



($J \sim 10$),

(Cu–O–Cu) = $94 - 90^\circ$ [56, 58].

1.2.3

LiCu₂O₂

LiCu₂O₂

[79, 80].

LiCu₂O₂

(\parallel)

()

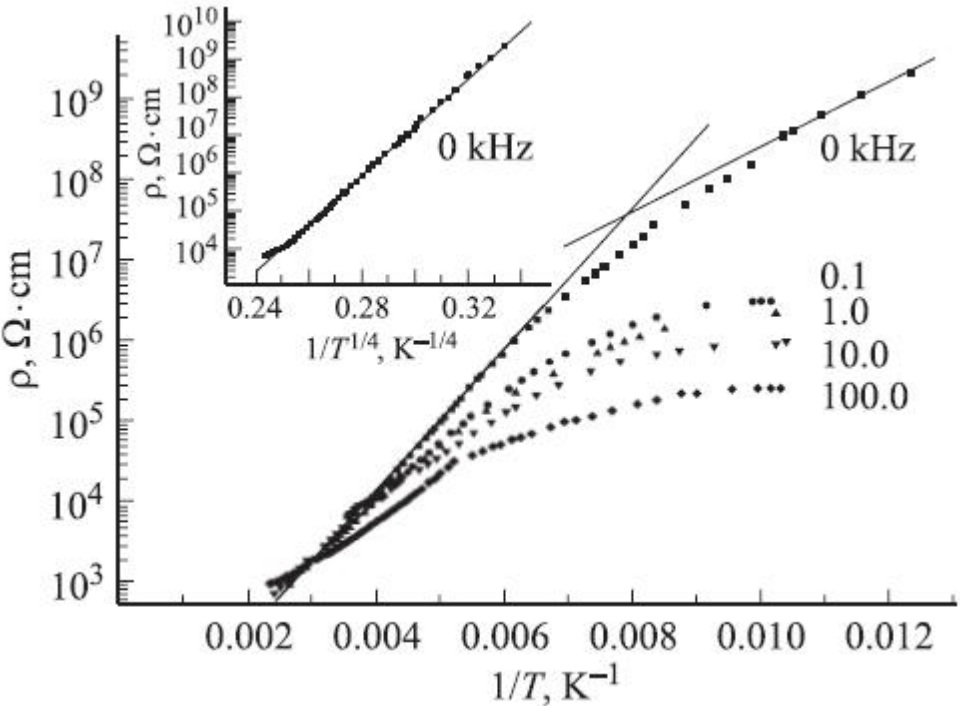
$\sim 10^3 \sim 10^2$ /cm

(.1.9).

LiCu₂O₂

$\lg -1/T$ (. 1.9).

$d(\lg)/d(1/T)$



1.9

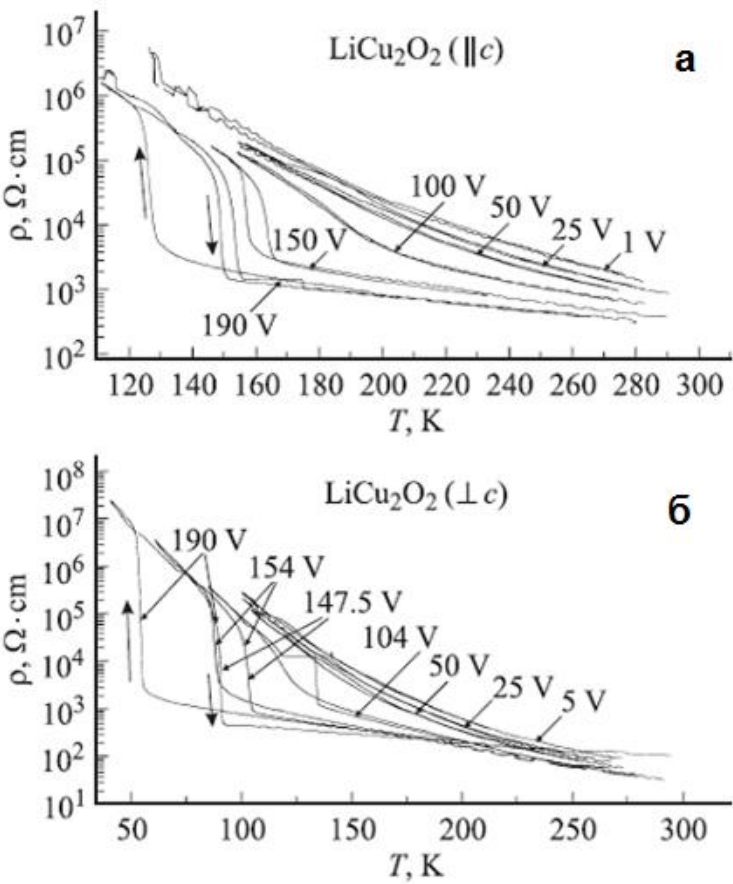
LiCu₂O₂

0.1, 1.0, 10.0, 100.0

$\lg -1/T$

(

$\lg -1/T^{1/4})/$



. 1.10

dc LiCu_2O_2 , ()

() U ().

80–250

.

LiCu_2O_2

[79].

-

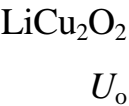
: () $\sim T^{1-s}$ (1.38).

s (1.38), 100 , $s \sim 0.2$

0.1–1.0 $s \sim 0.6$ 10–100 .

260 ,
 ,

, [79].



(. 1.10). ,



,
 , $U_o \approx 50$.

$\sim 10^3$.
 .

:

, .

ab. *c*

ab (. 1.10).



,

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.

 LiCu_2O_2 (~

30)

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2)

 LiCu_2O_2

S-

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3)

123.

4)

()

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1.4

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 LiCu_2O_2 .

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 LiCu_2O_2 ,

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 LiCu_2O_2

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2.

2.1

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 ,
 .
 - . -
 ,
 ,
 ,
 .[81]
 : 1)
 - 2)
 [81].

2.1.1 -

-
 .
 .
 ,
 « » (CuO – « . .», Li_2CO_3 « », ZnO « », AgNO_3
 « »).
 ,
 ,
 .
 .
 20÷100 50-100 .
 1 - 6 , 2 -
 50 / , -

().

~200 , , . , .

2.1.2

- () -2- () (.2.1) [82].

(5) (5000).

V

, .

: - 20÷3000

;

- $d/dx \sim 100$ / ;

- $V = 0,5 \div 33$ / .

= (10÷500) / .

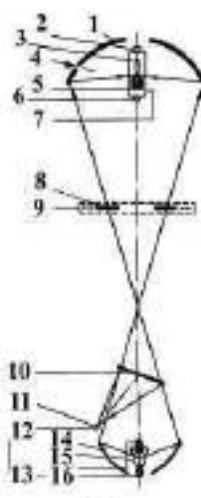
,

, - 12 10 .

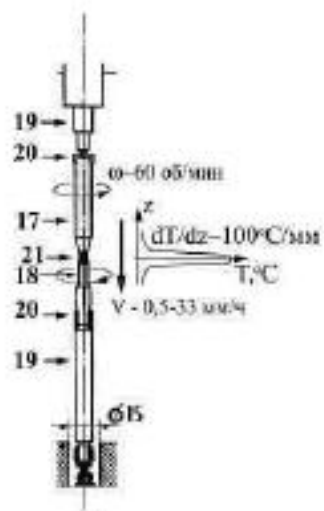


a)

2.1.



б)



в)

()

(,)

-2- . 1, 14 –

(), 15 –

, 3 –

, 11, 10 –

, 13, 16 12 –

,

, 8, 9 –

,

, 4 –

, 2, 6 –

, 17 –

, 18 –

, 19 –

, 20 –

, 21 –

.

(. 2.2),

LiCu₂O₂

.

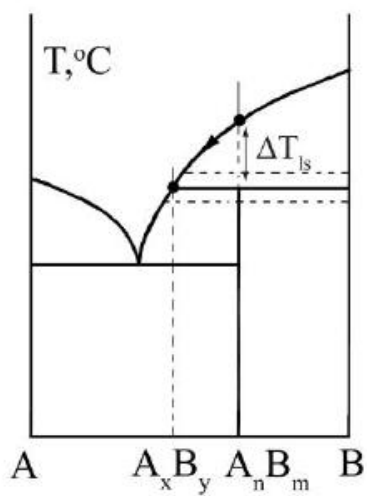
([83–86]),

 T_{ls} D

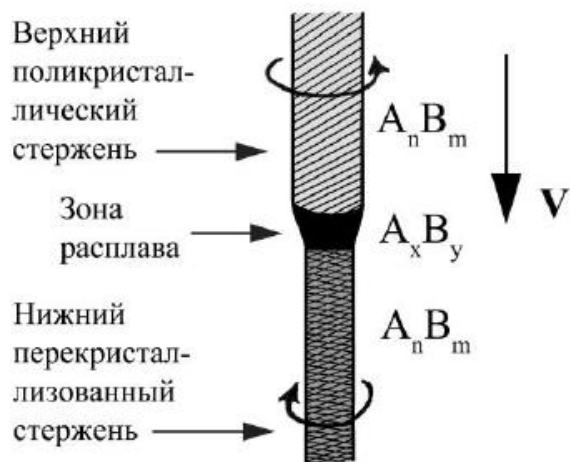
:

 $(d/dz)/V > T_{ls}/D$

(3.1)



2.2.



,

-

$A_n B_m$

$A_x B_y$

V_c (

)

,

$V > V_c$

.

(

)

,

V_c ,

,

(

dT/dz),

(

),

.

$LiCu_2O_2$

$LiCu_3O_3$

T_{ls}

~ 20

$V \sim 6$ / .

90

4

8

.

$\sim 100 \text{ } / ^2$,

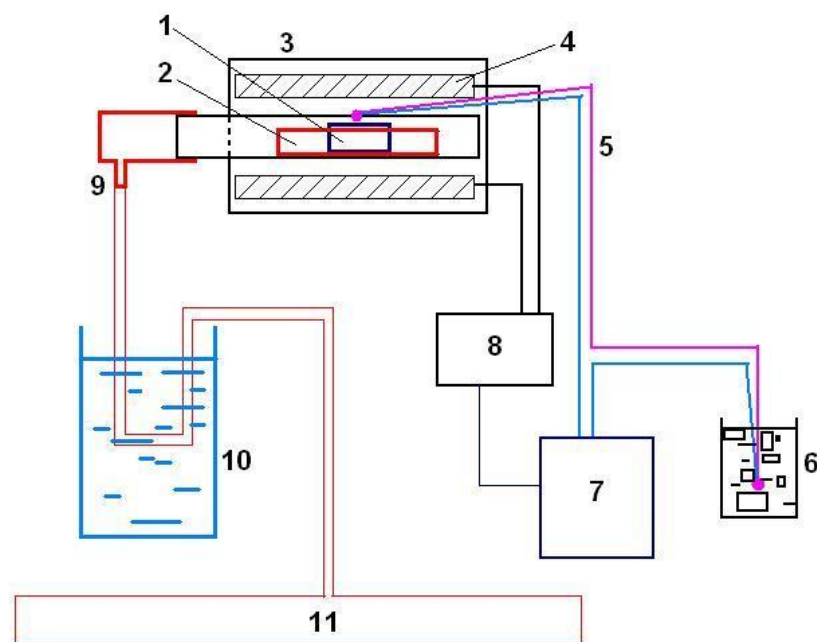
$3,0 \div 5,5 \text{ } /$,

2.2

- O

$\text{LiCu}_2\text{O}_{2+}$,

,
 $\sim 840^\circ\text{C}$
 ()



2.3.

: 1 – , 2 –

, 3 – , 4 –

, 5 – Cu-(Cu-Fe)

6, 7 – () – 101,

8 – – 101 (), 9 –

, 10 –

, 11 – .

.

. 2.3.

,

– 101 (,) ± 1°C,

.

.

()

.

O

c

.

2.3

2.3.1

d

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(

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.

.

()

,

d

-

:

$$2d/\sin = \quad (3.2)$$

-

.

()

-3,

-4 (. 2.4)[87]

: (CuK₁) = 1,54051, (CuK₂) = 1,54433.

ICDD

[88].

2

. 2.5.

2÷5 /

,

- 0,25÷0,50 / .

Ge (a =

5,6567(6)) Å

,

2

±0,02°.

,

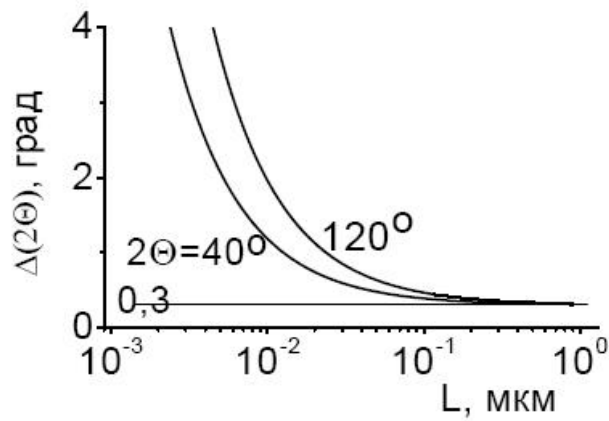
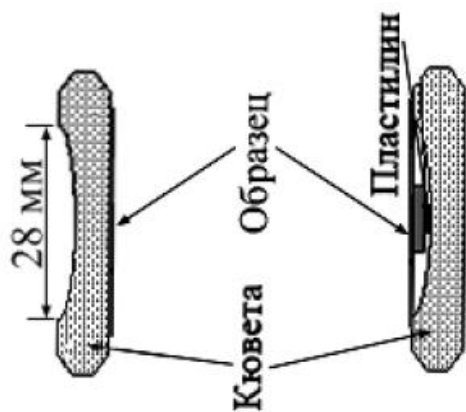
,

.



. 2.4.

-4



2.5) - () ()
 .)
 2 .

,
 d :
 $d/d = -ctg$. (3.3)

2.3.2

a, b, c

:

$$2d/\sin = , \tag{3.4}$$

$$1/d^2 = h^2/a^2 + k^2/b^2 + l^2/c^2 \tag{3.5}$$

h, k, l .

CELREF [89].

, .

2.3.3

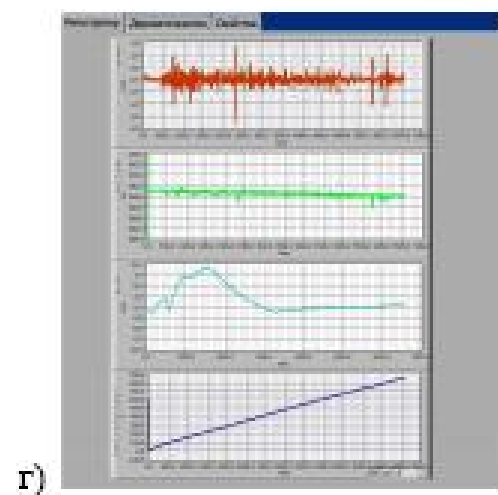
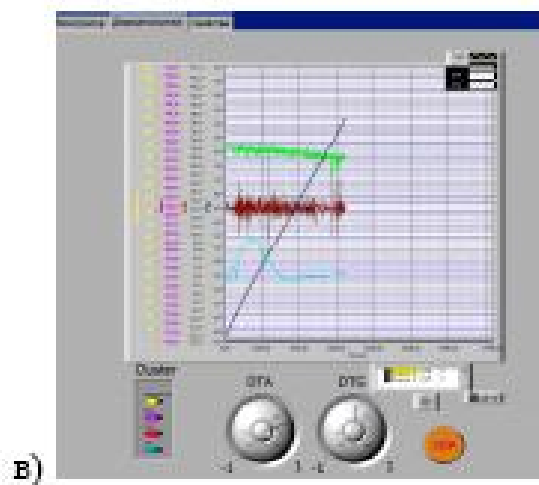
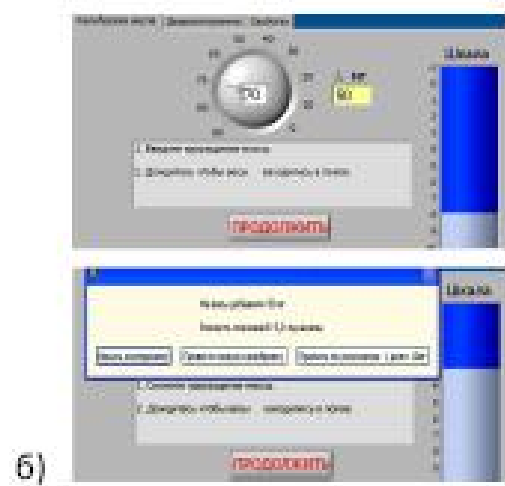
a, b c

- 2 .

2.4

()

()



2.6.

Q1500.vi

().

,

.

Q1500 D

F. Paulik, J. Paulic, L. Erdey.

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-

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2.5

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-

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2.7.

Au-Au:Fe,

0,1 ,

(.2.8)

-

.

5,26

.

-300

1

(

10^{-9})

Lock_in

amplifier,

10^{-9}

.

,

10^{-12} - 10^{-13}

.

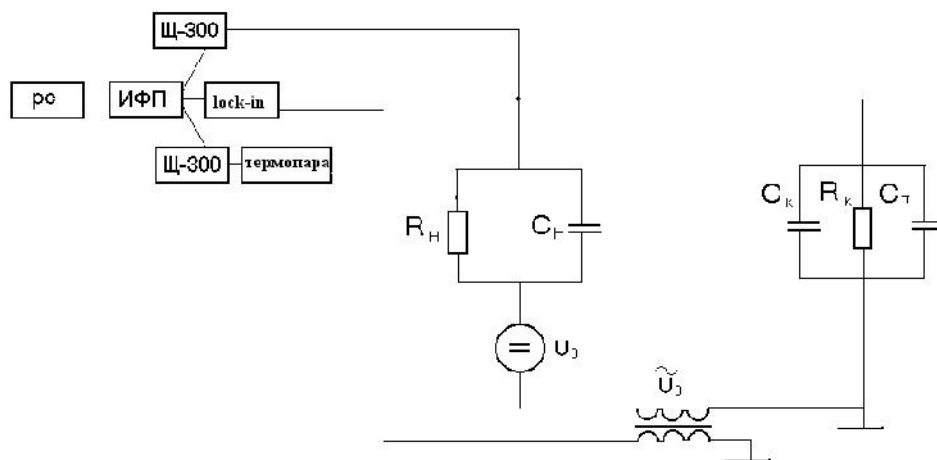
-

,

,

,

.



2.7.

, R C –, R C –, C –

.

 U_o $U_o \sim$, R

.

 $(0 < U_o < 250 \text{ В})$ $(1.74 \cdot 10^3 \text{ Ом}, U_o \sim = 100 \text{ В})$ $(R = 5,26 \text{ Ом})$

,

 (\dots)

.

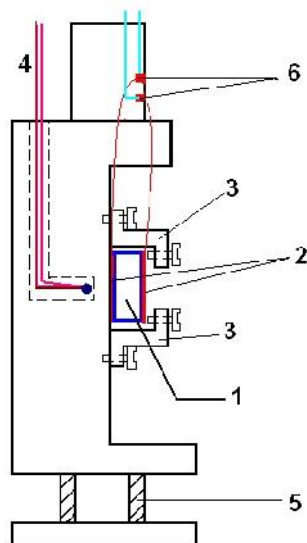
 $\dots ($ $)$

,

.

,

Labview (рис. 2.9).



2.8.

: 1 – , 2 –
In-Ga , 3 –
, 4 – Au-Au:Fe , 5 – , 6 –
.

7-78/1 10 (. 2.9).

, ,
, 0^0
7-78.

, LabView.

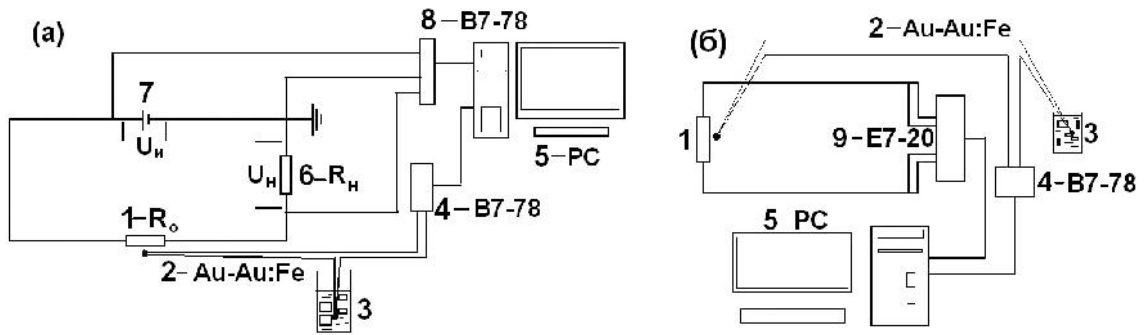
:

$$R_o = \frac{U - U}{U} R$$

(3.6)

U R – , U –
.

Au-Au:Fe.



2.9.

() () : 1 – ; 2 – Au-Au:Fe
 3; 4 8 – 7-78; 5 –
 ; 6 –
 , 9 – E7-20.

$$\dagger = \frac{l}{R_o \cdot S} \quad (3.7)$$

– , l S – .

7-20

0,1–100 .

. 2.9 . ,
 , 7-78.

LabView.

2.6

– 300 SQUID MPMS-XL-7 Quantum
 Design Inc (H = 20) ZFC (zero- field-cooled
 – –) FC (field-cooled –).

2.7

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().

“Orbis”

“EDAX” ()

Si(Li)

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Na.

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40

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60

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50

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():

-2 (. 2.10)

-

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-25

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20-50

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(25)

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(-4,



2.10. - " -2"

-5).

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 ,
 30 %.

$-2 \cdot 10^{-4} - 10^{-5}$ %.

-100,

.

3.

3.1

LiCu_2O_2
 $(\text{Li}_{1-x}\text{Ag}_x)\text{Cu}_2\text{O}_2$ $\text{Li}(\text{Cu}_{1-x}\text{Zn}_x)_2\text{O}_2$ 0 x 0,05 0 x
 0,12 .

3.1.1

-

 LiCu_2O_2

[51–54]

LiCu_2O_2 1163 – 1323 ,

-

. 3.1.

 CuO

« . »

 Li_2CO_3

« ».

 LiCu_2O_2 $x\text{CuO} \cdot (1-x)\text{Li}_2\text{CO}_3$ c 0,77 x 0,83

4

1393

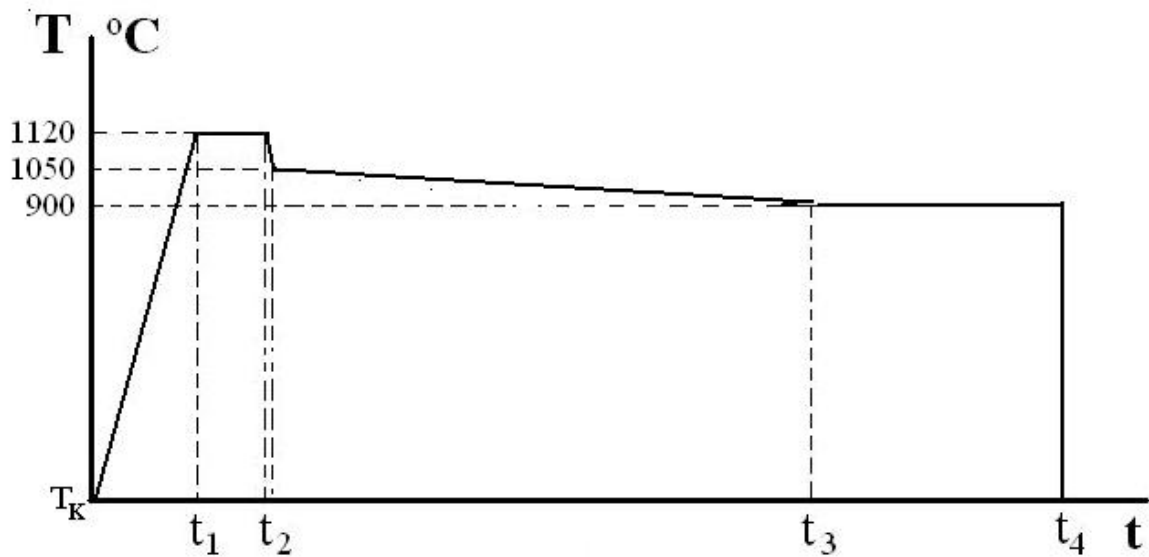
0,5 ,

.

1323 ,

2,0 ./

1173 ,



3.1

-

.



3.2 –

LiCu₂O₂.

1173 (20 - 24 °C),

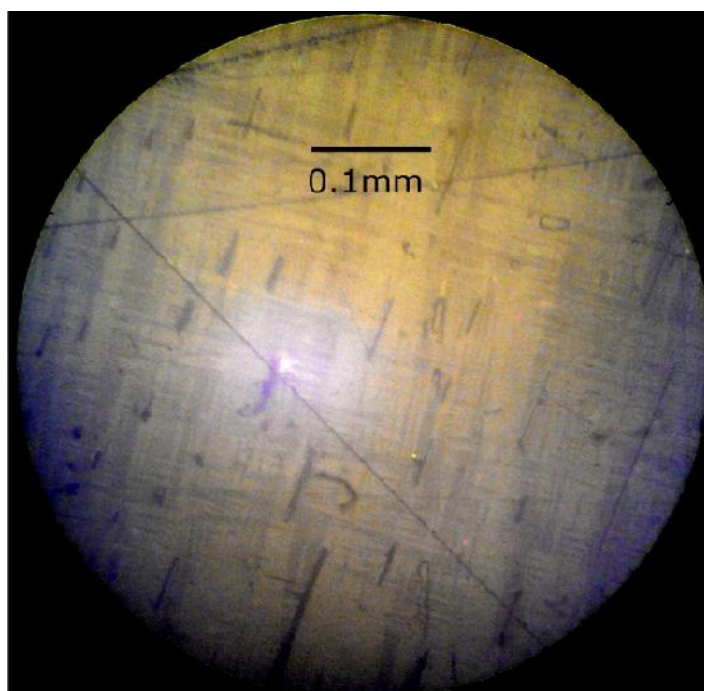
LiCu₂O₂

1393

1323

~1173

LiCu₂O₂ [3.2.1].LiCu₂O₂(0,5–4) × 10⁻³ (3.2).



3.3.

LiCu₂O₂,

(001),

(210) (2-10),

{001}

{210}.

LiCu₂O₂,

(120) (1-20).

(001)

(. 3.3).

5,2 / ³.

()

LiCu₂O₂

3.

3.1.2.

- LiCu_2O_2
 Ag , Zn
 Li_2CO_3 ,
 CuO , AgNO_3 ZnO « », « », « » « » ,
 $\text{Li}_2\text{CO}_3 \cdot 4(1-x)\text{CuO} \cdot 4x\text{AgNO}_3$,
 $\text{Li}_2\text{CO}_3 \cdot 4(1-x)\text{CuO} \cdot 4x\text{ZnO}$ - I $(1-x)\text{Li}_2\text{CO}_3 \cdot 2x\text{ZnO} \cdot 4\text{CuO}$ - II 0 x 0,5,
 LiCu_2O_2 .
 LCO ,
 $> 0,15$
 1 .

3.1.3.

 LiCu_2O_2 LiCu_2O_2

6

90 .



3.4.

1113

: 400

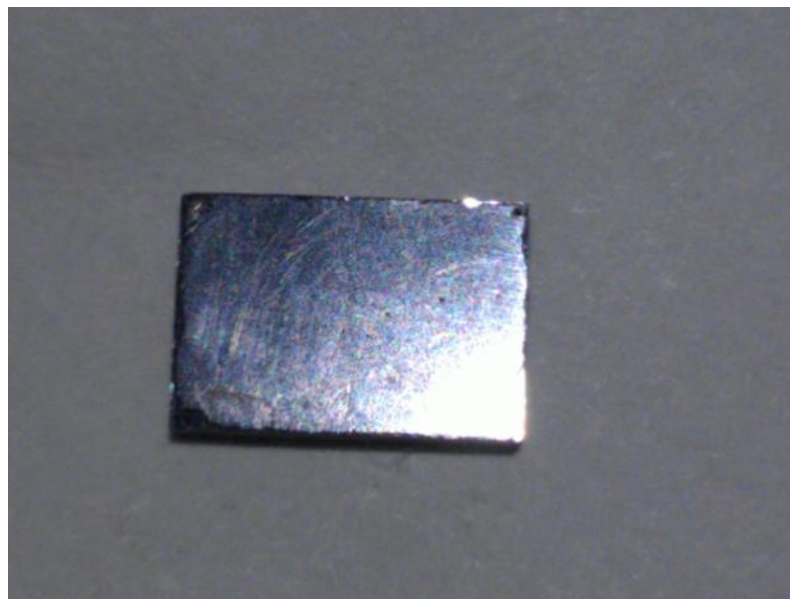
1 - 4 .

3.1

 LiCu_2O_2 **3.1.5.**

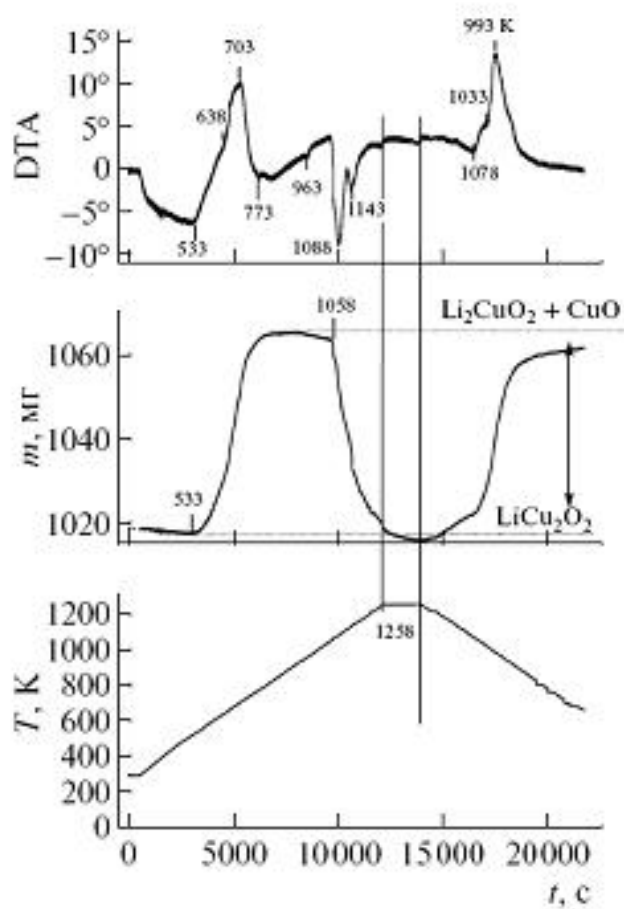
(001), (210) (2-10) (. 3.5).

In-Ga



3.5

.
 ,
 , -
 . ,
 .
3.2.
3.2.1
LiCu₂O₂:
 Q-1500D
 Pt-
 $T =$
 1200° - Al₂O₃.
 , LiCu₂O₂
 [90], $T = 553-773$
 LiCu₂O₂ Li₂CuO₂ CuO.
 (. 3.6) Cu⁺ Cu²⁺,
 LiCu²⁺Cu⁺O₂
 LiCu₂O₂ (683 °) + O₂↓ → LiCu₃O₃ (783 °) + O₂↓ + Li₂CuO₂ → Li₂CuO₂ + CuO + O₂
 1073–1223
 LiCu₃O₃, LiCu₂O₂,
 :
 Li₂CuO₂ (1108 °) + O₂↑ + CuO → LiCu₃O₃ (1163 °) + O₂↑ + Li₂CuO₂ + → LiCu₂O₂
 ,
 LiCu₂O₂ (1163 ° < T < 1323 °),
 , LiCu₂O₂



3.6.

 LiCu_2O_2 ,

(DTA –

, m –, T –, t –

).

DTA

1163–1323

 Li_2CuO_2

,

783 – 800

 LiCu_2O_2 Li_2CuO_2 CuO .

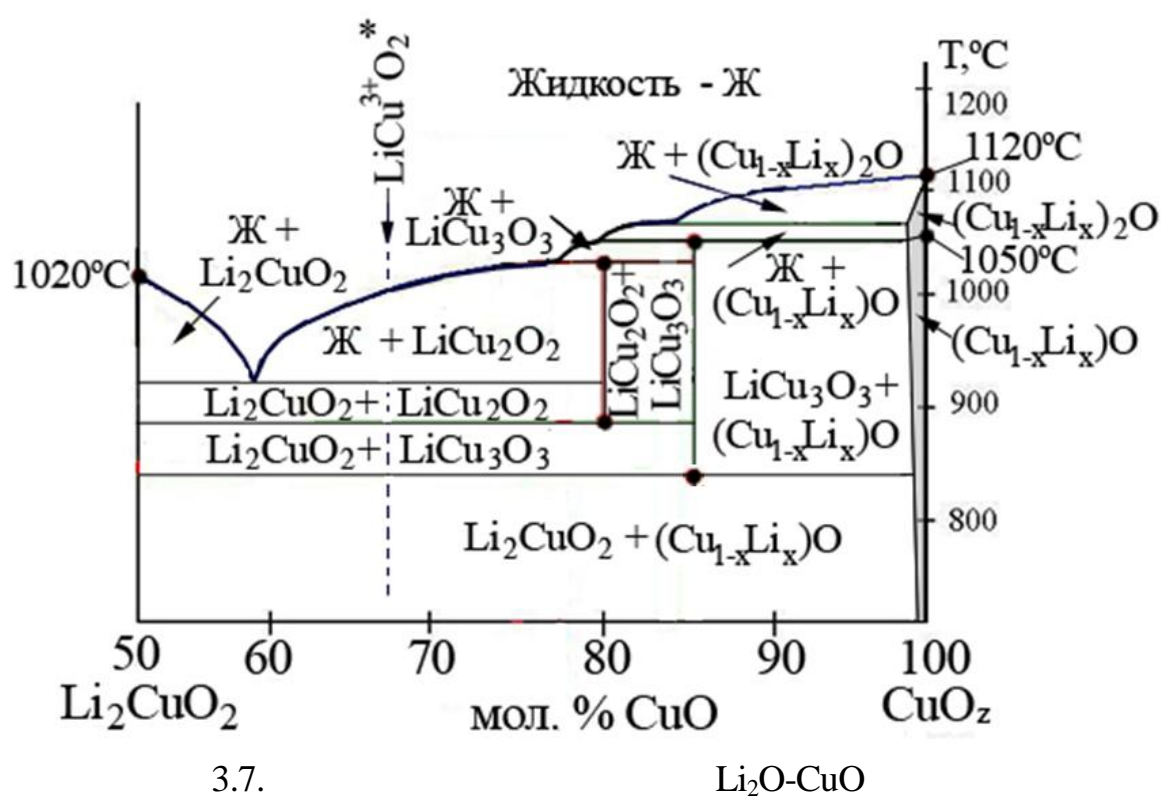
,

 LiCu_2O_2

1163 – 1373 .

1323

 LiCu_2O_2 .



3.7.

 $\text{Li}_2\text{O}-\text{CuO}$

(. 3.7).

: LiCu_2O_2 LiCu_3O_3 ,

(1323 1373)

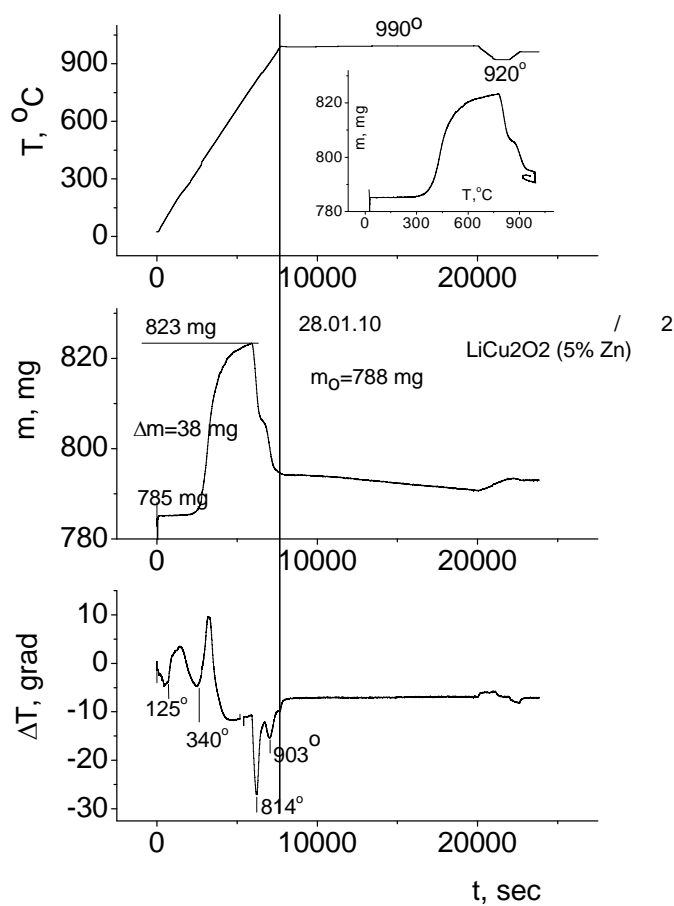
(1163 1113

).

1173 - 1323

 $\text{Li}(\text{Cu,Zn})_2\text{O}_2$, 5 .% Zn LiCu_2O_2 ; $\text{Li}(\text{Cu}_{0.95}\text{Zn}_{0.05})_2\text{O}_2$ (. 3.8) LiCu_2O_2 5

Zn



3.8.

Li(Zn_{0.05}Cu_{0.95})₂O₂:

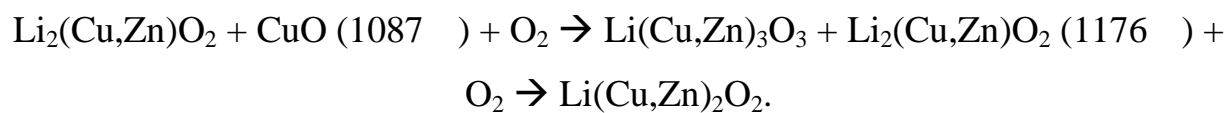
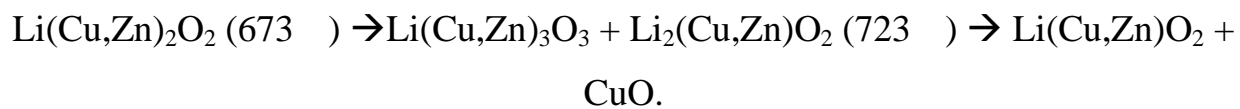
7,5 / ,

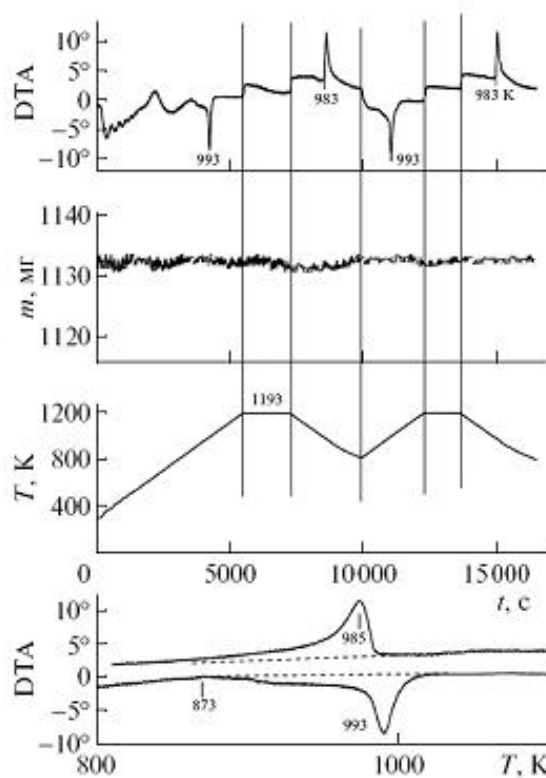
788,0 ,

(

1500 Al₂O₃)

.





3.9.

 LiCu_2O_2 ,

(DTA –

, m –, T –, t –).

DTA

3.2.2.

 LiCu_2O_2 LiCu_2O_2

(1320).

 $T_{PT} = 993$

,

(. 3.9).

,

.

DTA $T = 983$

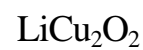
,

.

DTA

3.3.

3.3.1

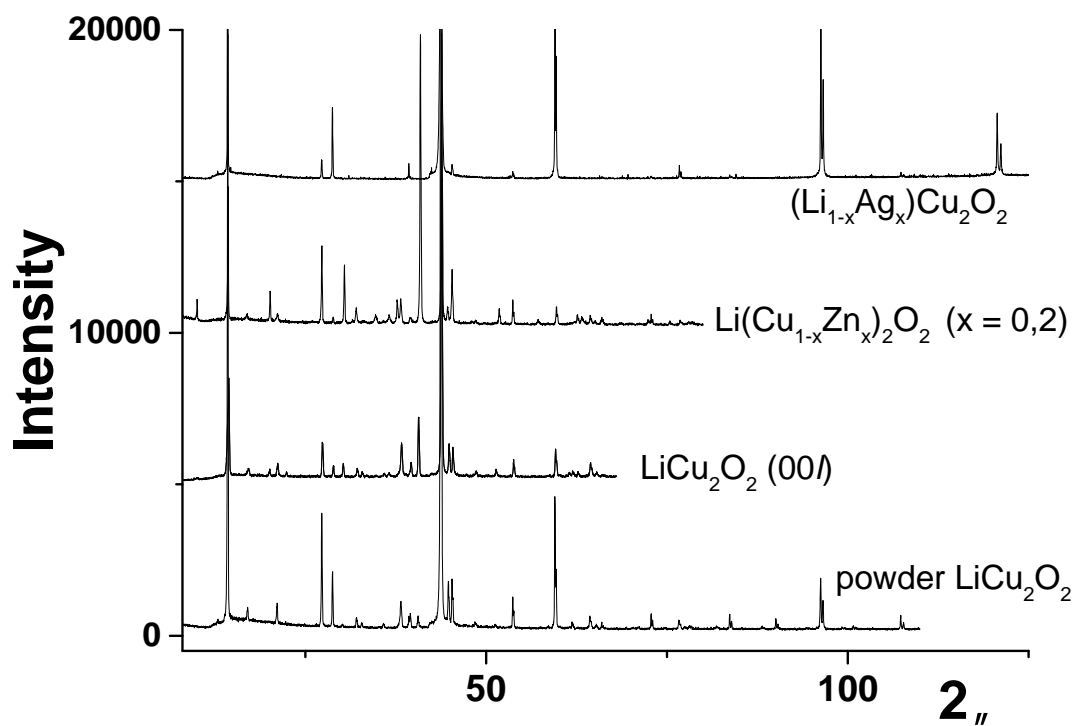
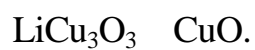


.

$$a = 5,726(2), \quad b = 2,858(1), \quad c = 12,410(2) \text{ \AA},$$

[52–54, 84] LiCu_2O_2 .

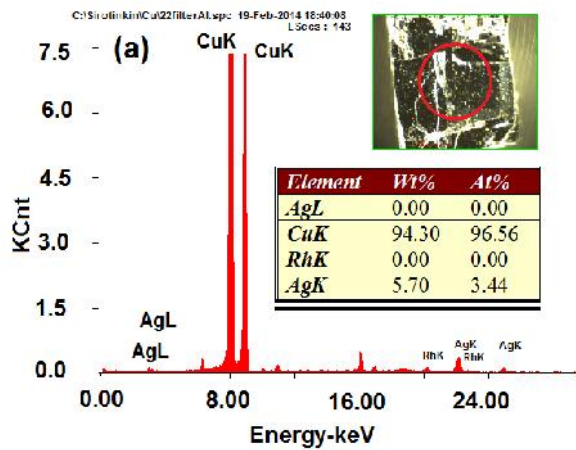
5%



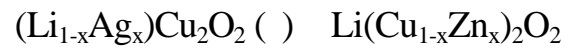
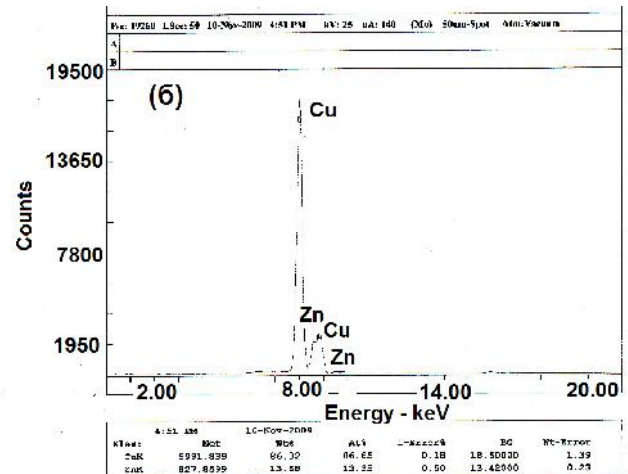
. 3.10.

,

-2 .



. 3.11.



().

, (400) (020),

-2 (. 3.3.1).

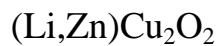
ZnO

Zn



Zn

Li



3.3.2

3.3.2.1

Li O,

. 3.11.

Ag



4 .%

Ag x = 0,25.

Zn
 $\text{Li}_2\text{CO}_3 \cdot 4(1-x)\text{CuO} \cdot 4x\text{ZnO} = 0,10 \quad 12 \quad .\% \quad ($
 $)$.

Li_2O_2 $4 \quad .\% \text{ Ag} \quad 12 \quad .\% \text{ Zn}$.

$(\text{Li}_{1-x}\text{Ag}_x)\text{Cu}_2\text{O}_2 \quad \text{Li}(\text{Cu}_{1-x}\text{Zn}_x)_2\text{O}_2 \quad 0 \quad x \quad 0,05 \quad 0 \quad x \quad 0,12$

3.3.2.2

—

3.2 LiCu_2O_2 () .			
	%	%	% , .
Li	4,0552	18,8013	20
Cu	74,2941	37,6456	40
O	21,5616	43,3606	40
C	0,0605	0,1620	
Al	0,0129	0,0154	
Na	0,0032	0,0045	
Ca	0,0043	0,0034	
K	0,0017	0,0014	
Mg	0,0008	0,0011	
Si	0,0009	0,0010	
S	0,0010	0,0010	
Cl	0,0007	0,0007	
Fe	0,0010	0,0006	
Cr	0,0005	0,0003	
Mn	0,0005	0,0003	
B	0,0001	0,0002	
P	0,0002	0,0002	

3.3 LiCu ₂ O ₂ ().			
	%	%	% ..
Li	4,0552	18,8013	20
Cu	74,2941	37,6456	40
O	21,5616	43,3606	40
C	0,0605	0,1620	
Al	0,0129	0,0154	
Na	0,0032	0,0045	
Ca	0,0043	0,0034	
K	0,0017	0,0014	
Mg	0,0008	0,0011	
Si	0,0009	0,0010	
S	0,0010	0,0010	
Cl	0,0007	0,0007	
Fe	0,0010	0,0006	
Cr	0,0005	0,0003	
Mn	0,0005	0,0003	
B	0,0001	0,0002	
P	0,0002	0,0002	

– ()

, [Cu]/[Li]

2 - ,

[O]/[Li] $2,2 - 2,3 = 2 +$

(. . 3.2 3.3).

= 0,2 - 0,3 O , ,

123 [91],

[92].

, O .

Al ,

•
•

3.3.3

Ag Zn

4 (CuK –). ,

[56] LiCu_2O_2 .

(204), (006), (210), (008),

(108), (400), (216), (200) (004), (006)

CELREF

a, b, c

(. 3.12).

Ag

a c

, *b*

, $x = 0,25$.

$x = 0,25$

.

,

,

,

LiCu_2O_2 4

.% Ag

.

,

[93] (. . 3.4),

,

,

LuCu_2O_2

Li^+

Cu^{2+} Ag^{2+} .

c

(c/c $x \sim 20\%$)

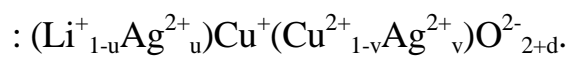
Ag^{2+} Li^+ Cu^{2+} (. . 3.4).

,

<p style="text-align: right;">. 3.4</p> <p style="text-align: right;">Li⁺, Cuⁿ⁺ Agⁿ⁺, n = 1, 2</p> <p style="text-align: right;">(. .) [93]</p> <p style="text-align: right;">,</p> <p>= [r(Agⁿ⁺)-r(C^{m+})]/r(C^{m+}), n, m = 1, 2, 3 (</p> <p style="text-align: right;">, * - . . =5</p> <p style="text-align: right;">. . = 4 . . =6, SQ - -</p> <p style="text-align: right;">).</p>					
	. . 2	. . 4	. . 5	. . 6	. . 8
Li ⁺ (1s ²)		0,590	0,675 *	0,76	0,92
Cu ⁺ (3d ¹⁰)	0,46	0,60	0,685 *	0,77	
Cu ²⁺ (3d ⁹)		0,57	0,65	0,73	
Cu ³⁺ (3d ⁸)				0,54	
Ag ⁺ (4d ¹⁰)	0,67	1,00 1,02 SQ	1,09	1,15	1,28
Ag ²⁺ (4d ⁹)		0,79 SQ	0,865 *	0,94	
Ag ³⁺ (4d ⁸)		0,67 SQ	0,705 *	0,74	
(Ag ⁺ -C ⁿ⁺) C ⁿ⁺ =Li ⁺ Cu ⁺ Cu ²⁺ Cu ³⁺	0,46	0,69 0,67 0,75	0,615 0,59 0,68	0,51 0,49 0,57 1,13	0,39
(Ag ²⁺ -C ⁿ⁺) C ⁿ⁺ =Li ⁺ Cu ⁺ Cu ²⁺ Cu ³⁺		0,34 0,32 0,39	0,28 0,26 0,33	0,24 0,22 0,29 0,74	
(Ag ³⁺ -C ⁿ⁺) C ⁿ⁺ =Li ⁺ Cu ⁺ Cu ²⁺		0,135 0,12 0,175	0,04 0,03 0,08	-0,03 -0,04 +0,01	

Ag²⁺ Li⁺, -

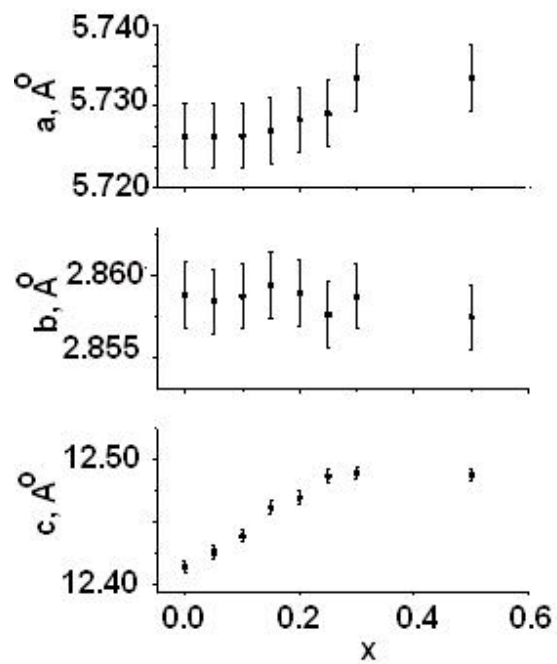
1) Cu¹⁺



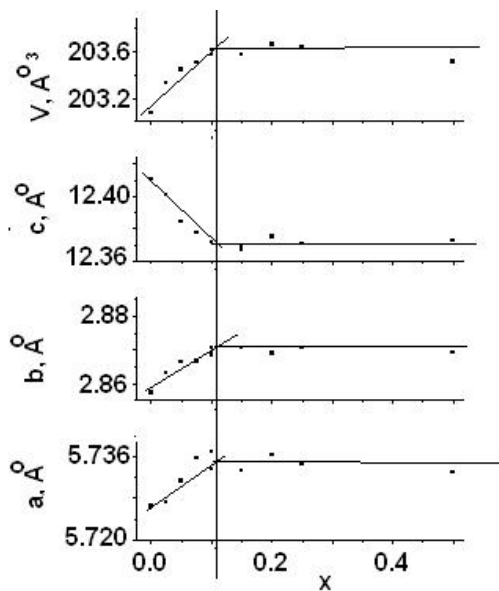
Zn

a, b c

= 0 - 0,12,



3.12.

Li(Cu_{1-x}Ag_x)₂O₂,

3.13.

 a, b, c

V

LiCu₂O₂Li(Cu_{1-x}Zn_x)O₂.

> 0,12

(. 3.13).

3.3.4

,

O ,

c (,

123 [91]) [79],

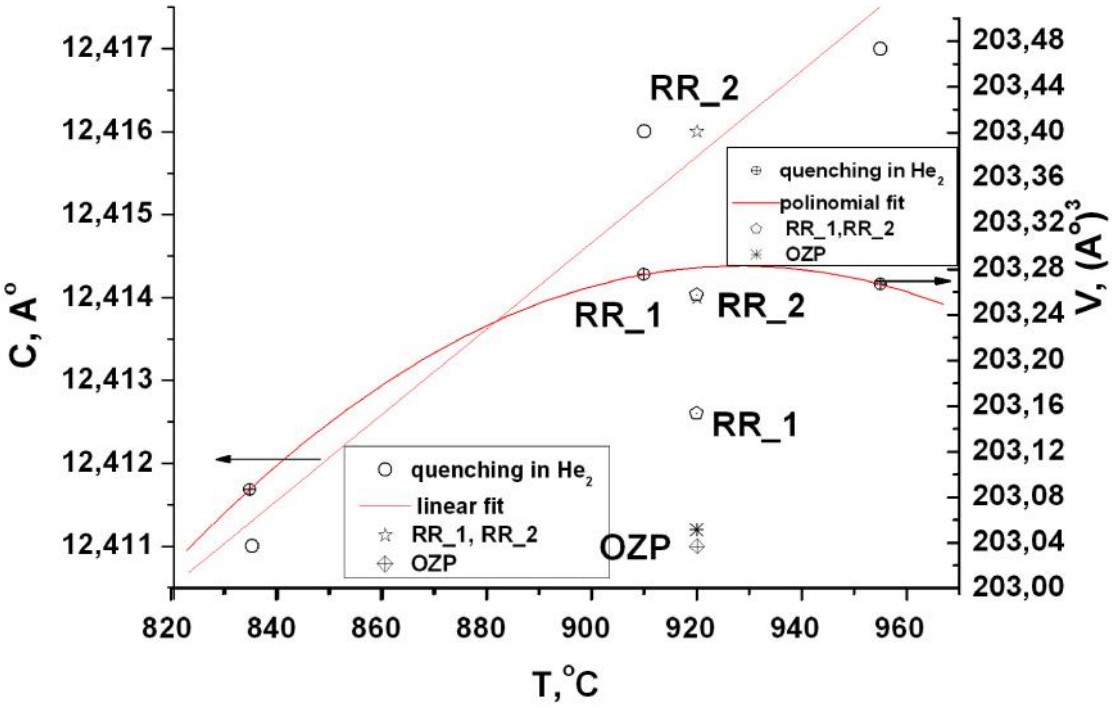
$p(\text{O}_2) = 0,21$, O

. 3.14

V

(³) c () ,

:



3.14.

O

LiCu_2O_2 .

<p>3.5 , (crys_) – , (crys) – .</p>		
A4	12,415(4)	12,412(1) 12,413(5) (crys_1) 12,412(1) (crys_2)
N3	12,418(1)	12,410(1)
W2	12,414(6)	12,412(6)

He₂. $\sim 1\%$ [95].

$_{-1} \quad _{-2},$

$$\sim 1193 \quad , \quad ($$

c ,

$$(\quad . \quad . 3.1)$$

. 3.5.

$$, \quad (A4,$$

N3),

(001) LiCu₃O₃,

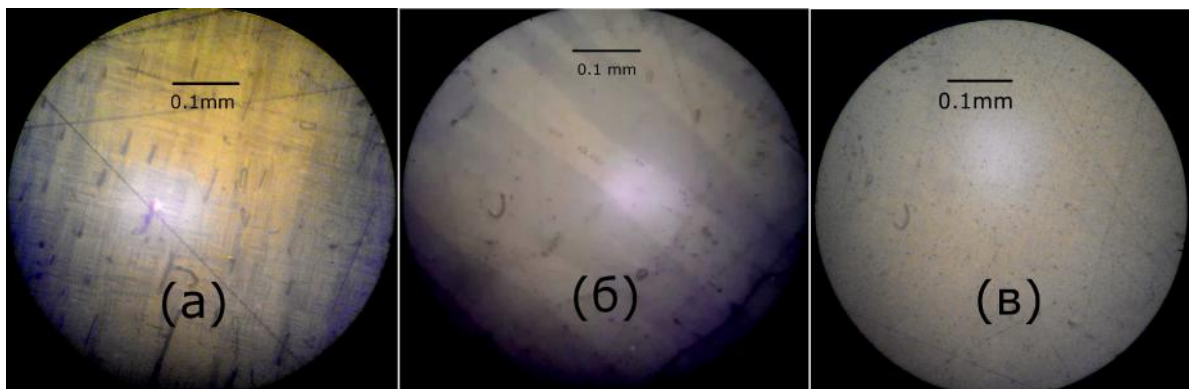
.

,

$$, \quad , \quad (001) \textit{ab-}$$

(3.15),

, [210] [2-10]



3.15.

(a)

()

().

(3.15).

ab

,

,

,

(0012)

2 .

().

:

(3.3).

W2,

2

,

(001) *ab*-

.

O

,

,

.

3.3.5

. 3.16

CuK₁ CuK₂,

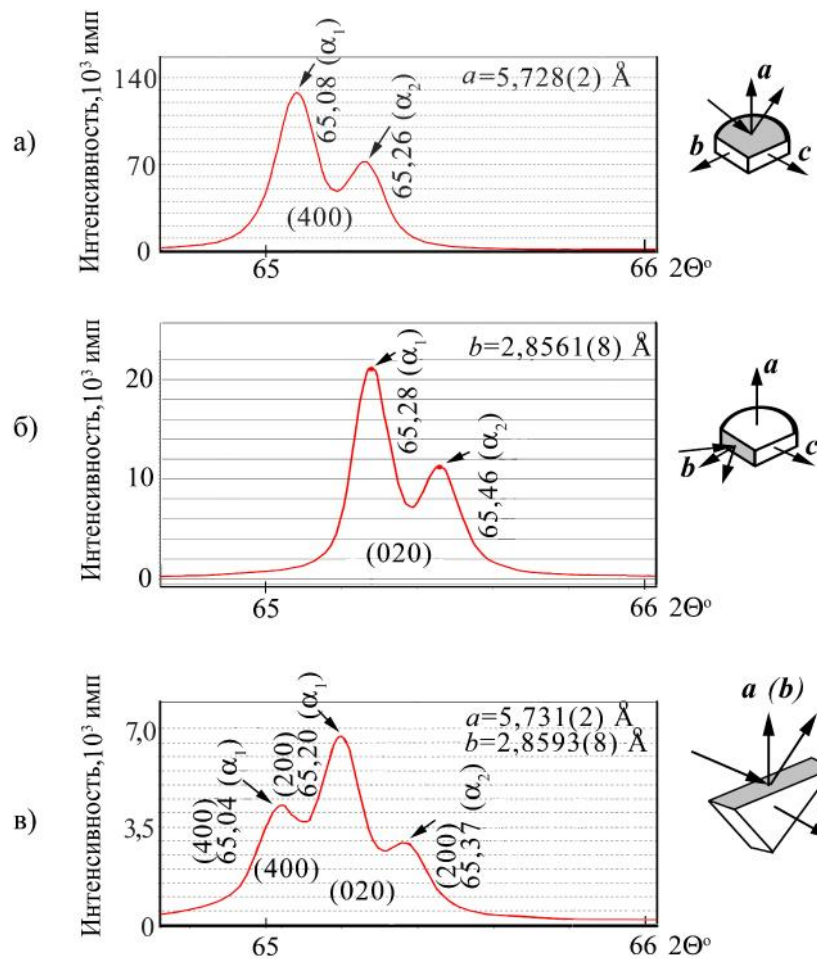
(

(400) – , (020) –) ((400) (020)

–). ,

1⁻ 2⁻

(400) (020),



3.16.

,

-2

LiCu₂O₂,

(400) (020)

(

–),),

–).

3.4.

LiCu_2O_2

3.4.1

LiCu_2O_2

$\text{LiCu}_2\text{O}_{2+}$,

:

(1, 2)

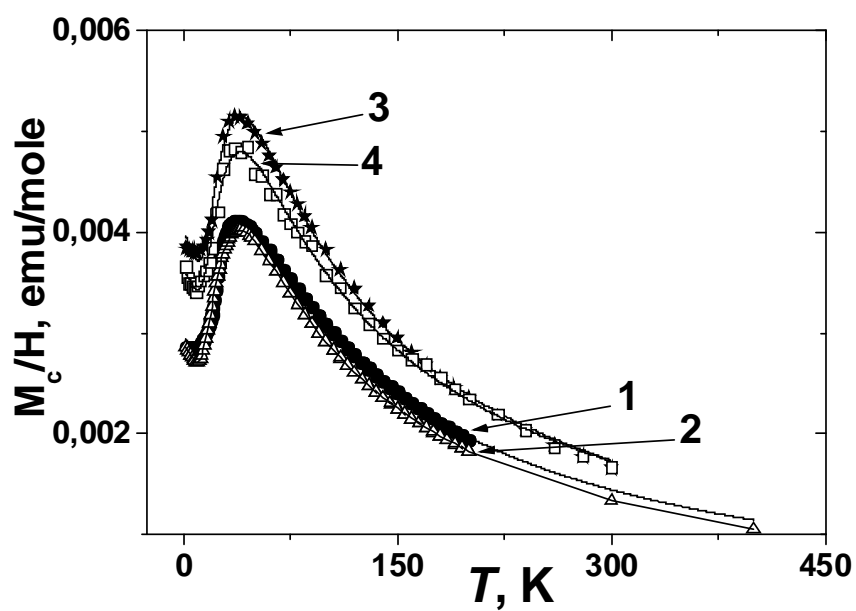
(1, 2).

$() = M/H$,

$(H = 10 \text{ Oe})$,

40 K,

- (Fig. 3.17).



3.17.

():

||

$H_{DC} = 10 \text{ Oe}$,

1

(1 2)

1

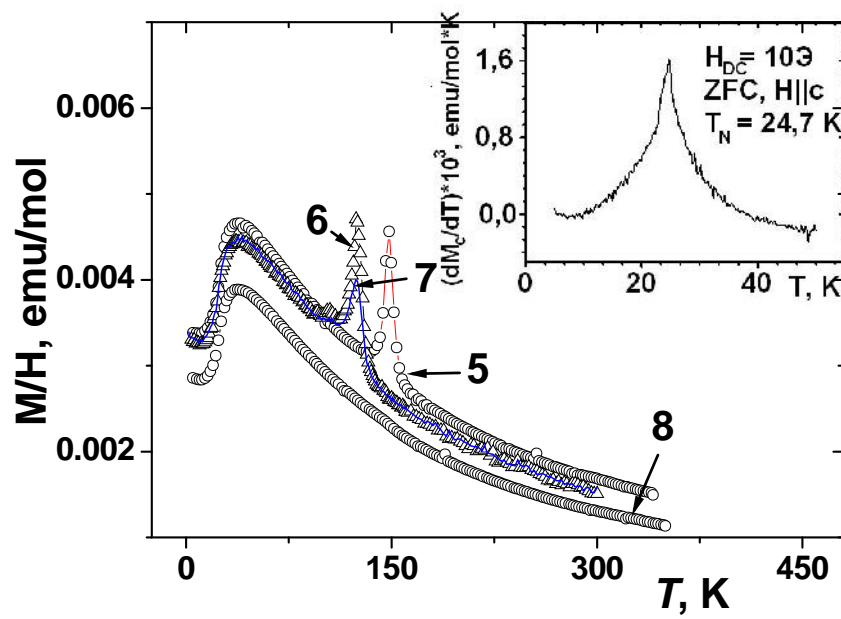
$f = 110$

$h_{ac} = 2 \text{ Oe}$ (3, 4).

1, Cu^{1+} , Cu^{2+}
 $S = 1/2$.

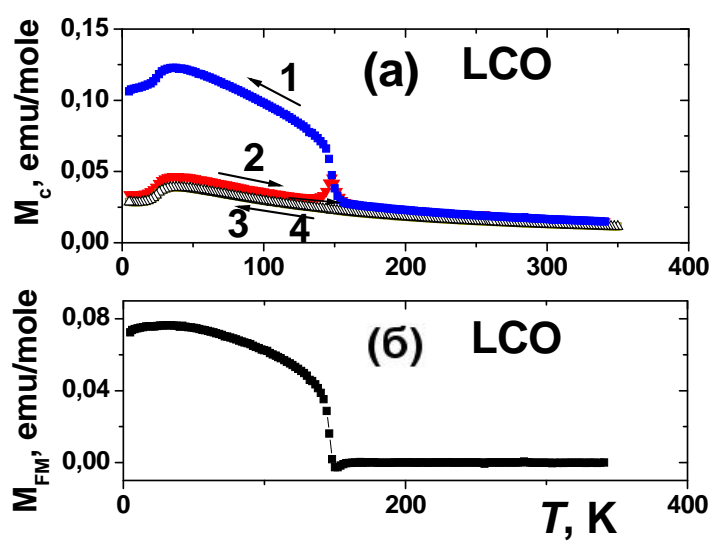
(T, H) .

O,



3.18.
 (T) ,
 5, 8 – $H_{\text{DC}} = 10$, $H \parallel c$,
 7 – $H_{\text{DC}} = 10$, $H \parallel b$,
 8 – $H_{\text{DC}} = 10$, $H \parallel c$,
 2, $T_N = 24.7$,
 LiCu_2O_2 , c .
 ZFC:
 6 – $H_{\text{DC}} = 0.5$, $H \parallel c$,
 1, 5, 6, 7, $dM_c(T)/dT$

(T)
 $\mu_{DC} = 10$ 1 (. 3.17). < 400 K,
O , () (),
.
.
CuO₂ .
 ,
O CuO₆, uO₄ .
u²⁺ -
(uO₄)⁺, +1.
() .
 , - O .
 , [92] , La₂CuO₄₊
180 - 400 .
. 3.18 (T) , $(H = 10 \text{ O})$.
 ,
 . (T) 148 124,7
2 2, ,
 $(T) \quad H \quad b \quad 2 \text{ (} 8)$
 ,
 .
(FC, ZFC).
 dM/dT 2,
 $N=24.7$
LiCu₂O₂.
O
 $(H = 10 \text{ O})$ =150



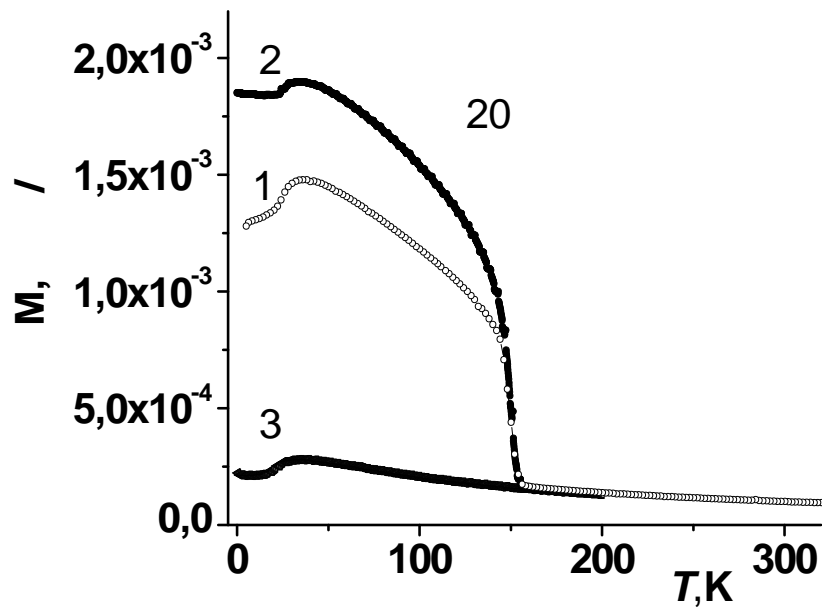
3.19. LiCu_2O_2 ()
 $(H \parallel c, H_{\text{DC}} = 10 \text{ Oe})$: (1) FC; (2)
 ZFC, $H \parallel b, H_{\text{DC}} = 10 \text{ Oe}$: (3)
 ZFC; (4) FC.)
 $M_{\text{FM}} = M_{\text{FC}} - M_{\text{ZFC}}$ ()
 2.

$\text{LiCu}_2\text{O}_{2+x}$
 (. 3.18)

3.4.2

LiCu_2O_2

, $\text{Li}_2\text{O} \cdot 4(1-x)\text{CuO} \cdot 4\text{AgNO}_3$ c
 $0 \leq x \leq 0,5$,
 $M(T)$ 5 – 300 (H
 $= 20 \text{ Oe}$), c ZFC (
) FC (). ZFC FC $M(T)$
 $T = 37 \text{ K}$ (. 3.20),
 $T_{c1} \quad T_{c2}$



3.20.

$\text{Li}(\text{Cu}_{1-x}\text{Ag}_x)_2\text{O}_2$ $x = 0$ (1), 0,05 (2) 0,15 (3)
20 .

$dM(T)/dT$.

$M(T)$

LiCu_2O_2 [68, 73, 94–97].

F

$M(T)$

$x = 0,05$,

,

$T_3 = 150$,

[69,

85, 90].

T_3

ZFC

FC.

Ag ($x > 0,05$)

$T_{c3} = 150$

(. 3.20).

Li^{1+}

Cu^{2+}

Ag^{2+} .

LiCu_2O_2 ,

3.5

LiCu₂O₂

3.5.1

LiCu₂O₂

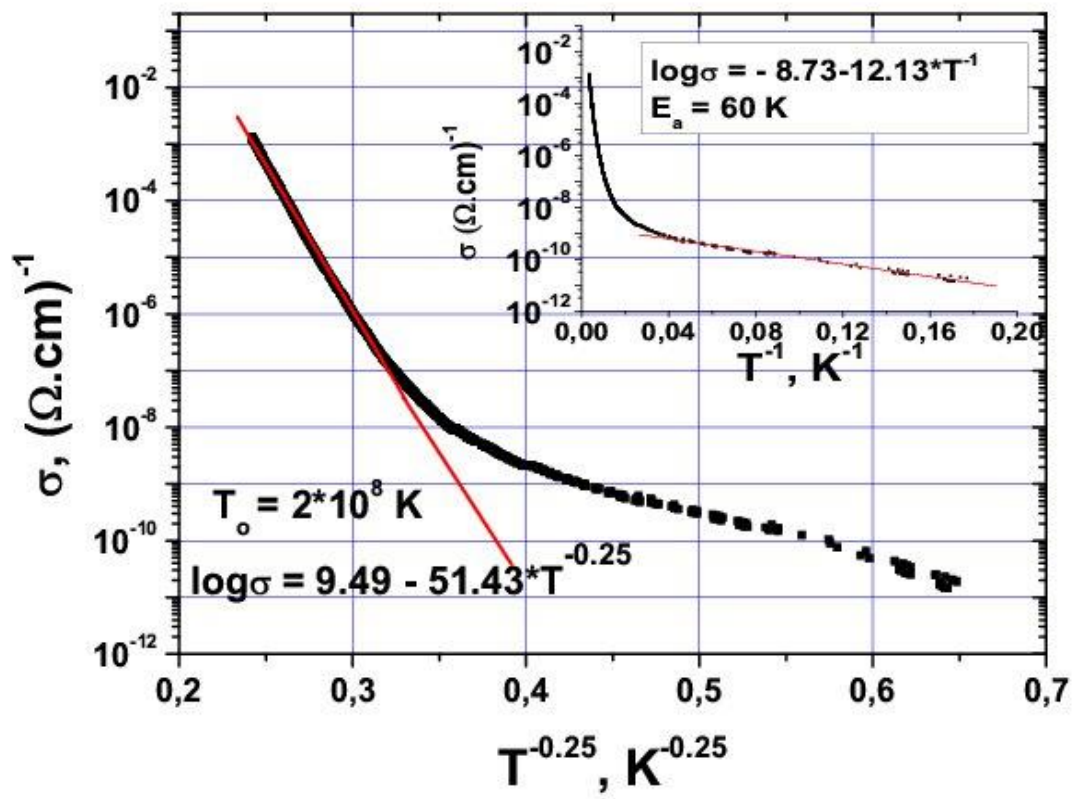
$$5 \times 5 \times 2 \times 10^{-3},$$

$$a \times b$$

3.21

$\sigma(T=0)$ 1. 3.21

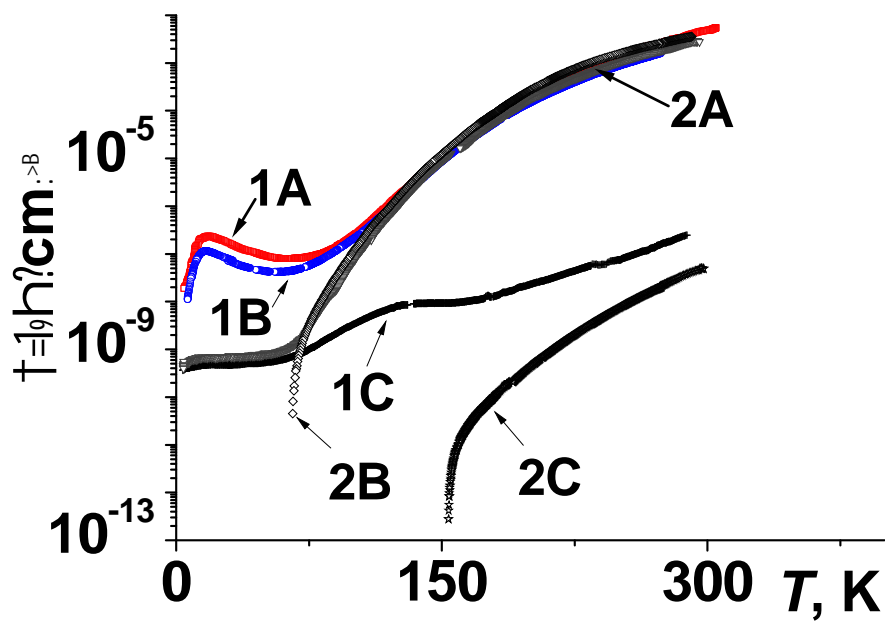
: $T^{-0.25}$ T^{-1} ().



3.21.

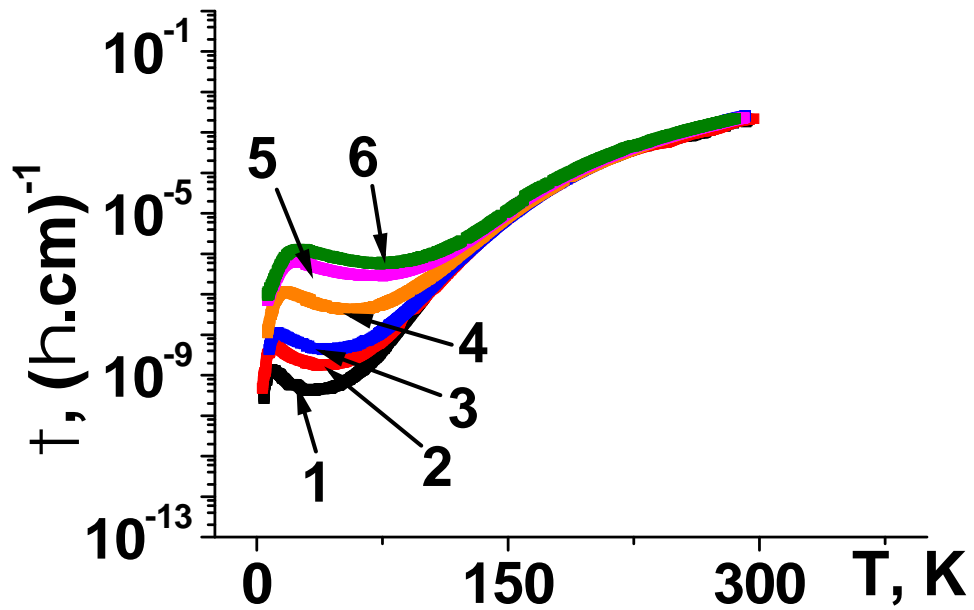
$\sigma(T=0)$ LiCu₂O₂ : $T^{-0.25}$

T^{-1} ().



3.22. AC DC (1 , 1 1
10 ~1 B,
a, b, c; 2 - 4,5 0,3 , 2
2 - ~10 , M1).

,
.
300 – 100
 $\epsilon_0 \sim 10^6 - 10^8$. 300–360
 T^1 $a \sim 0,35 - 0,44$.
25 DC :
 $a \sim 5 - 6$,
.
. 3.22
AC DC a, b c
1.
[79].



3.23. $\tau(T)$ $b(1, 2, 3, 4, 5, 6)$ 0,1; 0,5; 1; 10; 50 100 .

DC

$T \sim 295$ K

c

$ab.$

b

$a.$

$$E_D = -d(\ln \tau)/d(1/T),$$

E_D 0,15; 0,12 0,1 0,3, 107 150 , ~ 200

$$(\ln \tau \sim 1/T)$$

$$(\ln \tau \sim T^{0.25}).$$

E_D

b

c

~ 200

0,158

0,246

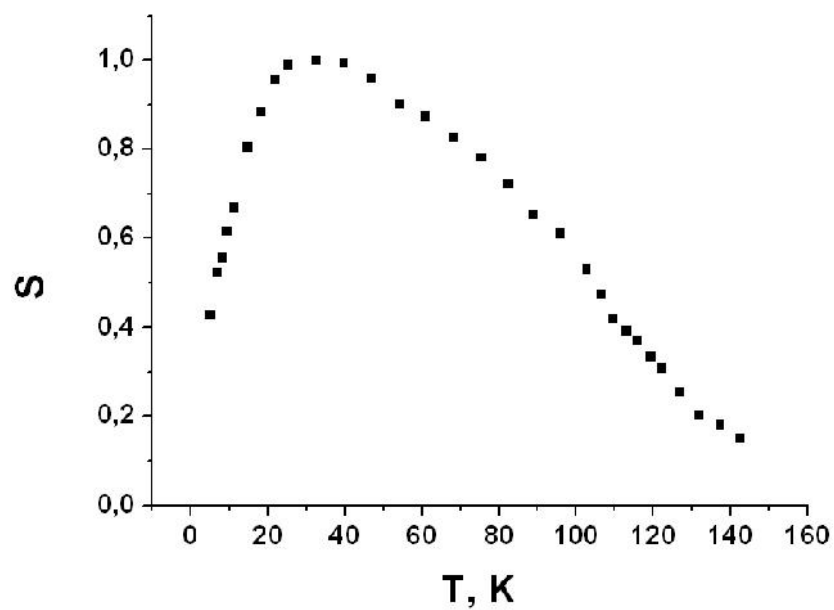
1

1

. 3.22.

ab

$T <$



3.24.

$s()$

1

b.

,

a b

16.6 15.7 ,

. *c*

— ~134 .

,

.

()

,

(1.38).

. 3.23

($T,)$,

M1

b

. ,

~100

80

,

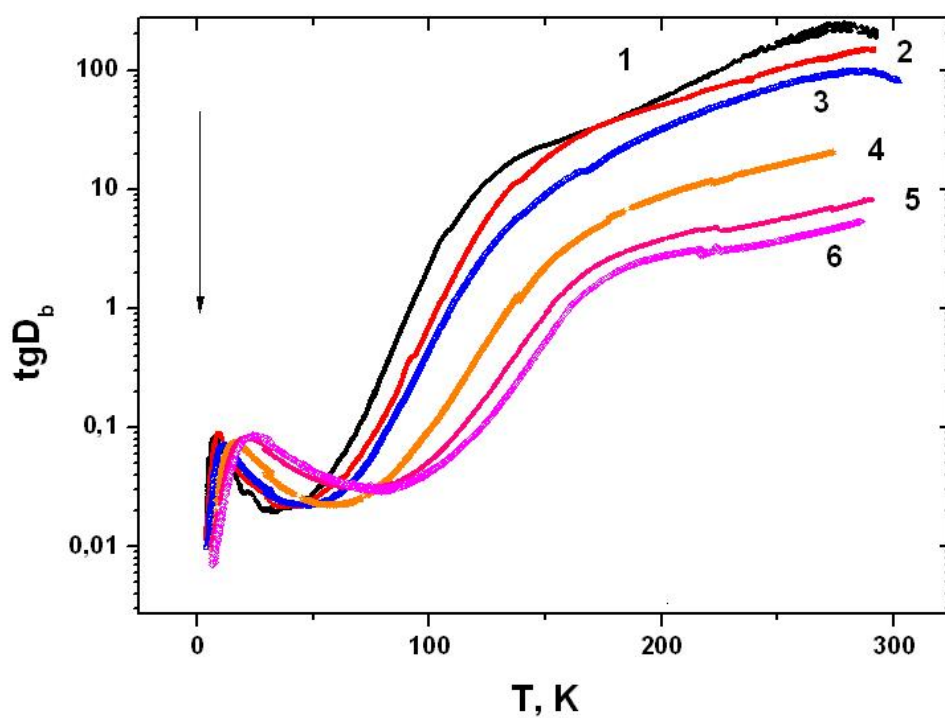
80

,

(1.38): ($T,) = T^n s$) $s \sim 1$,

()

.



3.25. $\text{tg}(T)$ b
 0,1; 0,5; 1;10; 50; 100 (1, 2, 3, 4, 5, 6).

(1.38) 0.1–100

()

$s()$ (. 3.24)

,

[47, 51]. $s(T)$,

(1.38) .3.23

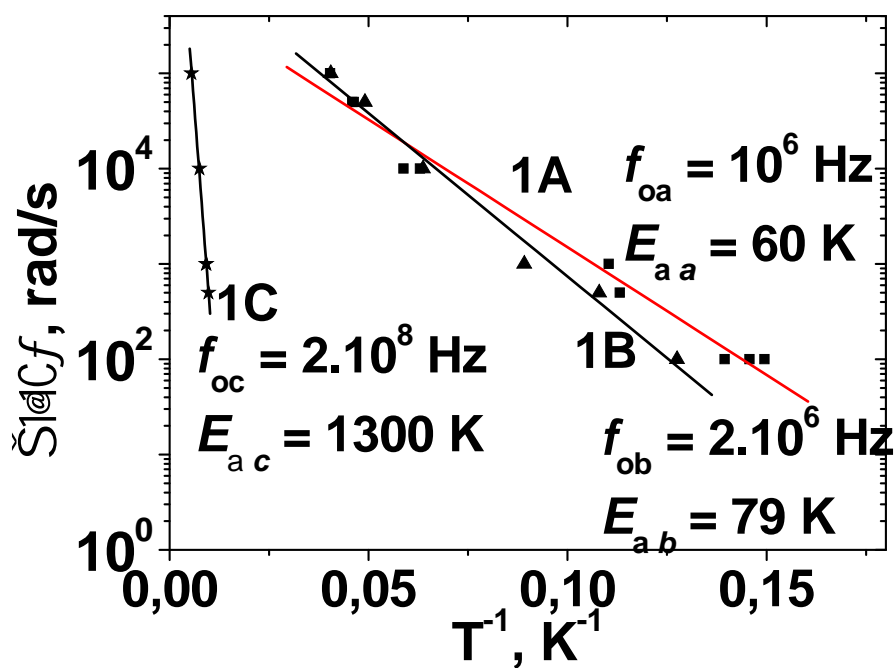
4,5 – 80

$s(T)$

.3.24. $s(T)$ (= 30

), [34, 47, 51, 98].

30



3.26.

, M1

a, b, c .

100 250 ()

, .

LiCu_2O_2

$\text{tg } (T)$

4 – 295

F : 0.1 – 100 .

$\text{tg } (T) \quad T = T_{\text{max}}$.

. 3.25

$\text{tg } (T) \quad b$

.

(1.39) [47].

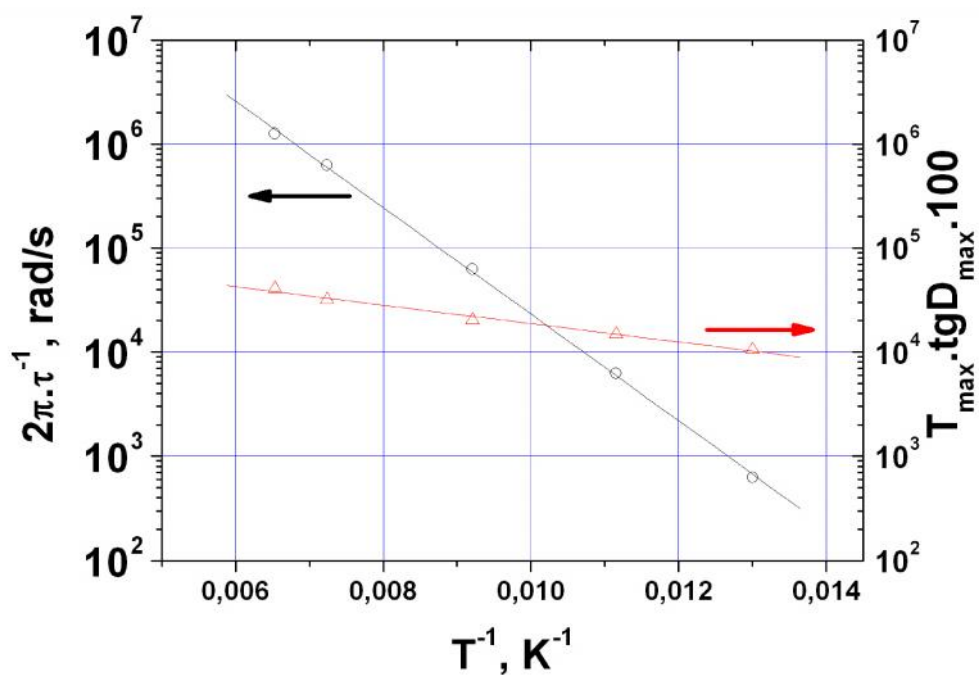
,

$\text{tg } (T) \quad a, b, c,$

,

$\lg f - 1/T_{\text{max}}$,

(. 3.26).



3.27. $f_{\max}(T) = (\text{tg}_{\max} \cdot T_{\max})(T)$ M1 **b.**

$$(\omega = 1/2 f),$$

$$= f_{oi} \exp(E_{ai}/kT), \quad i = a, b, c. \quad f_{oi},$$

$$\sim 10^6$$

$$a \quad b \quad 2 \cdot 10^8 - \quad f_{oa,b} \quad E_{a,a,b}$$

CuO₂

$$(b) \quad (c) [66]. \quad c$$

$$f_o \quad (c).$$

$$< 30 \text{ K} \quad > 30 \text{ K} -$$

[52]:

$$\text{tg}_{\max}(T^{-1}) \sim [(\text{tg}_{\max}^{-1}) \cdot \exp(-Q_a/T)].$$

$$.3.27 \quad 2 / (T^{-1})$$

$$T_{\max} \cdot \text{tg}_{\max}(T^{-1}) \quad E_a, f_o \quad Q_a.$$

$$E_H = 2Q_a.$$

3.5.2

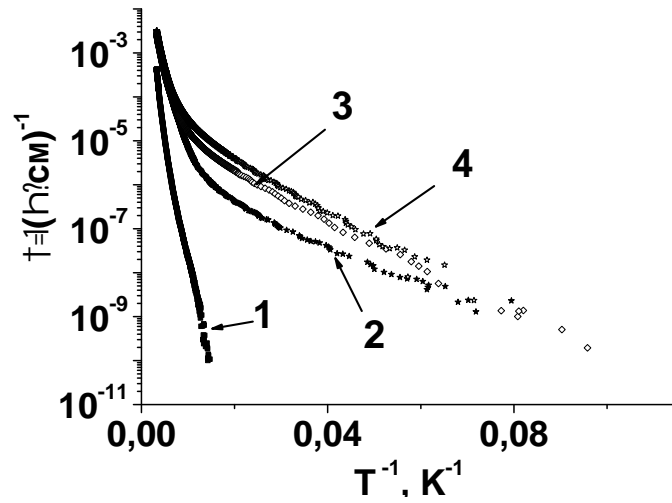
LiCu₂O₂

3.5.2.1

A4.

DC

A4



3.28. DC

A4

c (

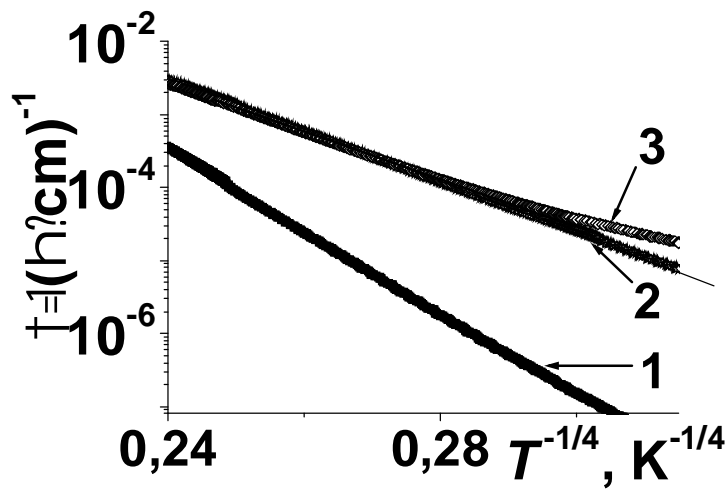
1),

(2)

1113

(3 4)

T^1



3.29. DC

A4 (

)

c (

1),

(2)

1113

(3)

$T^{1/4}$.

ab 20%, c
(. 3.28).
1113 40
— ()
.
 c ,
, ,

(1.12) T_o .

T_o .
, ,
, ,
.

(~24) —
.
:

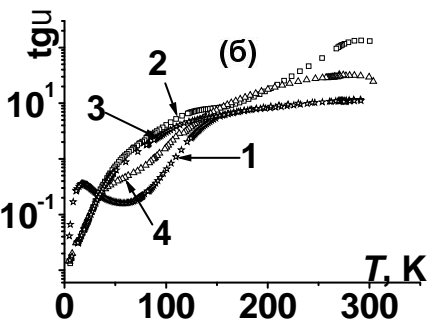
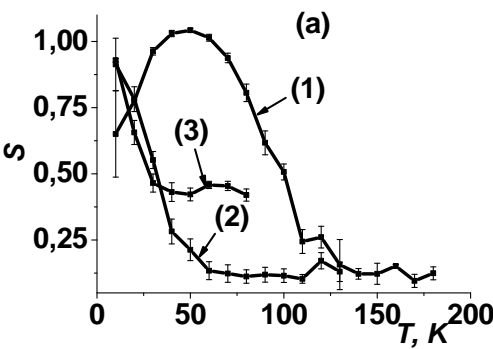
,
.

10, 20, 50 100 .
 $s(T)$,

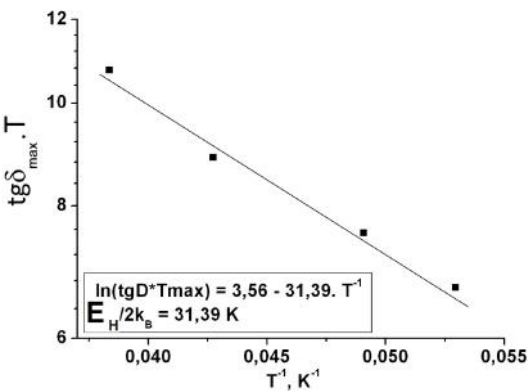
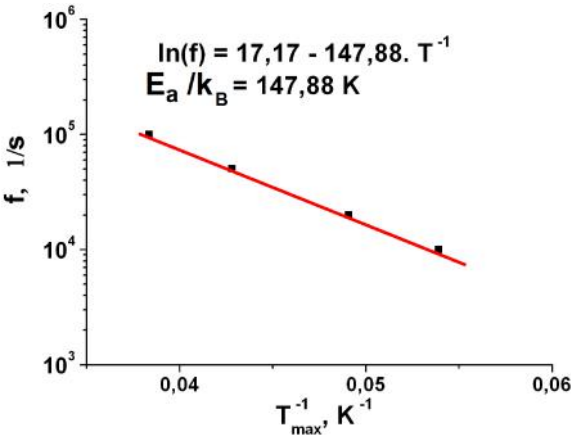
, $s(T)$
(. 3.30a).

$\text{tg } (T)$ A4
20 , ;
200 .

(. 3.30).
,
,



3.30. a) $s(T)$ *ab* A4 (1),
(2) (3);) $tg(T)$ *ab*
10 .



3.31. $f (tg_{max} \cdot T_m)$ A4 *ab* .

, $\gg 1$.

ab $N = 24$ (

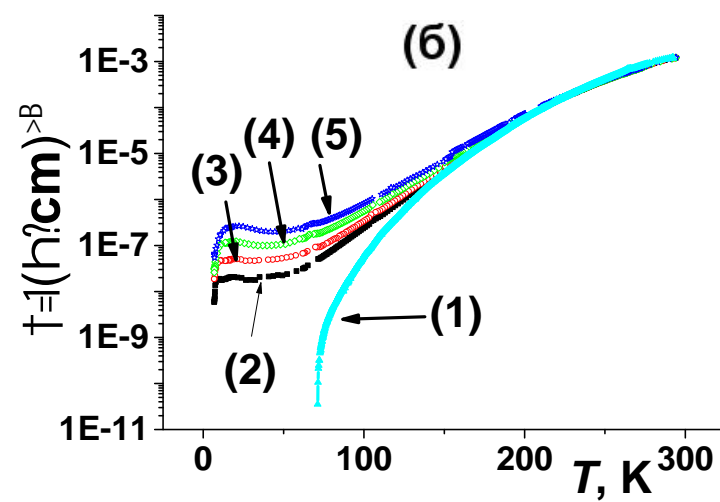
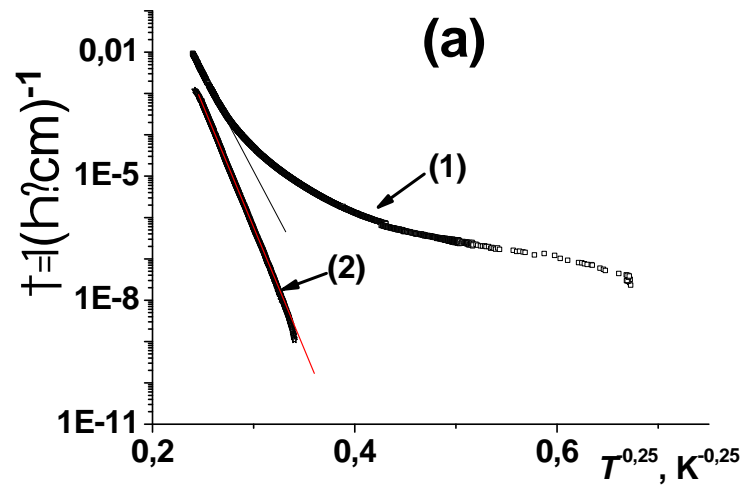
) $a/k_B = 147,9$ $E_H/2k_B = 31,4$ (. 3.31).

() (. 3.30) .

3.5.2.2

N3.

2, 2 400 2 1 .
DC AC
DC



3.32. a) DC N3 *ab* ()
(2) .) DC (1) AC (2,3,4,5
10, 20, 50, 100) N3 .

~ 300 $b = 10^{-2} (\dots)^{-1} \sim a \sim 10^* c,$
 ab $c E_a = 0,16$
 $T_o = 0,97.10^8$ $\sim 260 \dots$ $T \sim 230$ $E_a = 0,175$ $T_o =$
 $3,60.10^8$ $(\dots 3.32a).$

AC

\dots $24 \dots$

$ab,$

$(\dots 3.32 \dots).$

\dots

ab

$\sim 200 \dots$

\dots

N3

«

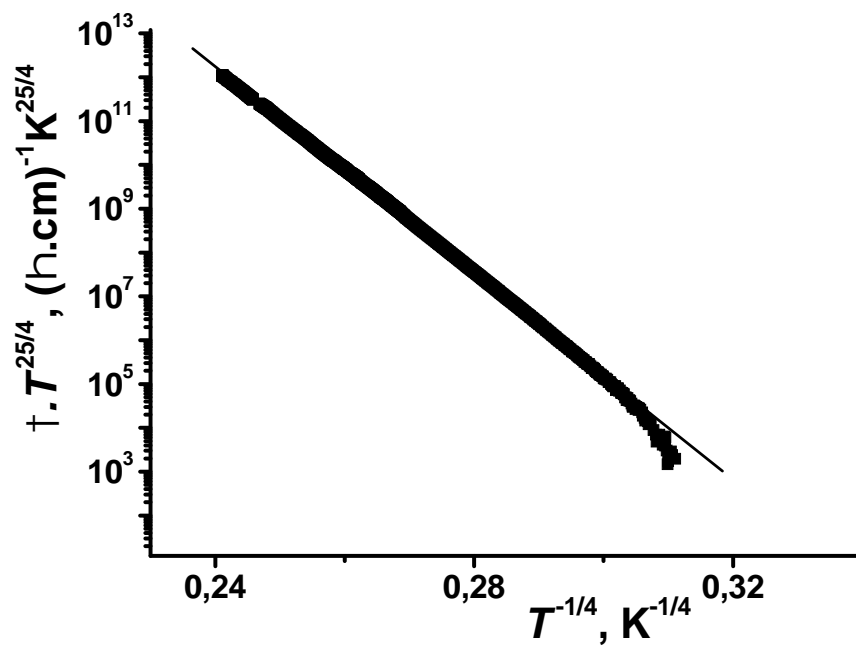
»

1.

A4,

\dots

\dots

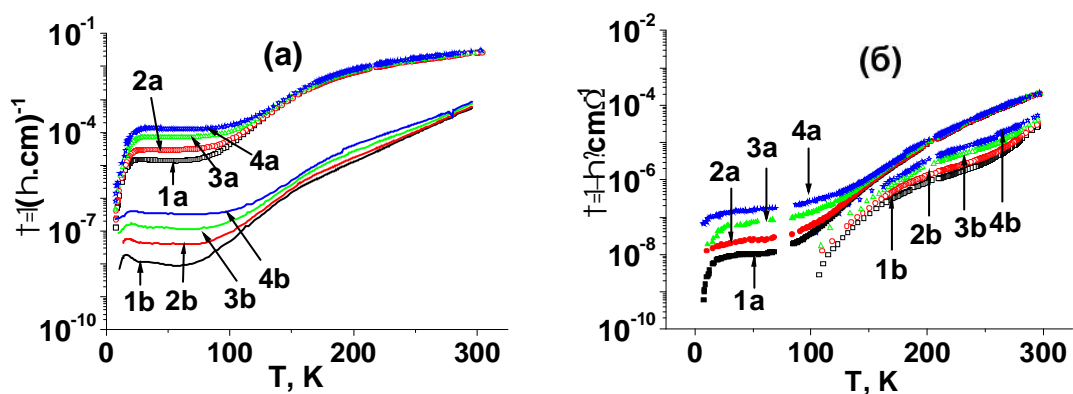


3.33. DC

W2

ab

$\dots T^{25/4} (T^{-1/4}).$



3.34. AC

W2 () ,

(a)

 ab () (c a)

(

b)

10, 20, 50 100 (

1, 2, 3, 4

).

,

-

.

3.5.2.3

W2

W2 $T \sim 300$

:

$$\tau_b = 2 \cdot 10^{-2} \text{ (s)}^{-1} \sim \tau_a \sim 10^2 \text{ (s)}^{-1}$$

DC

.

$$ab \sim 10^{-10} \text{ (s)}^{-1}$$

 ~ 50 ,

-

 ~ 100 .

,

(

).

,

 ab

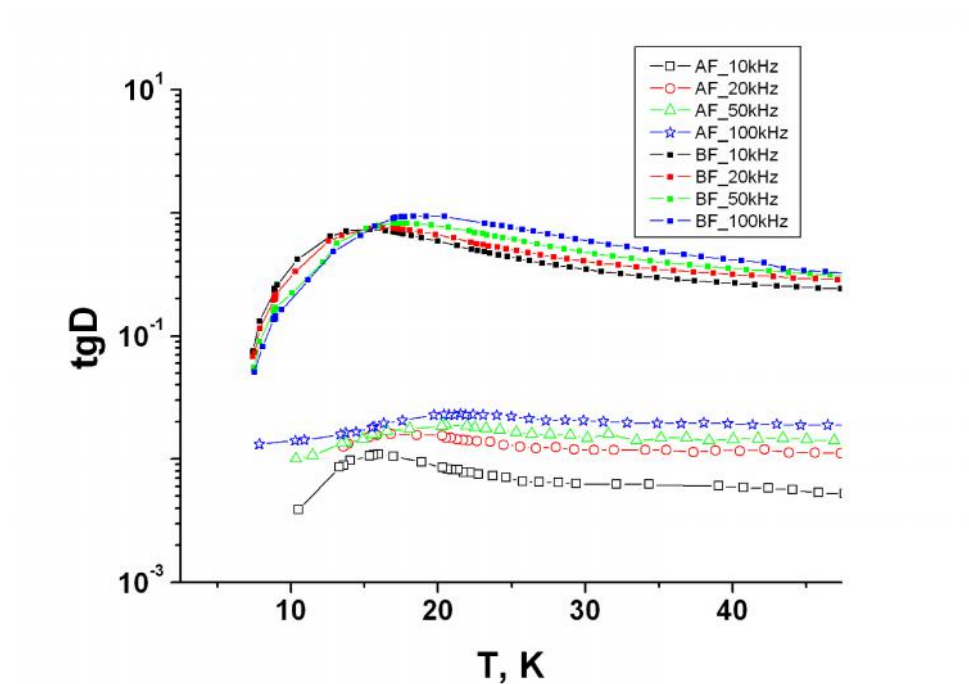
$$m = 25/4 \quad T_0 = 6,75 \cdot 10^7 \text{ (K)} \quad (3.33).$$

[35]

,

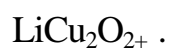
,

 $\text{Li}^+ \quad \text{Cu}^{2+}$



3.35 $\text{tg}(\) \text{W2 } ab$

,



AC

. ,

$< N$ (. 3.34a).

.

,

,

DC

ab,

,

,

,

,

.

DC

,

.

~300

~100

ab,

.

AC

:

tg_{max} ,

[35, 52, 99],

ab (. 3.35).

ab

.

3.5.3

 LiCu_2O_2

3.5.3.1

 $\text{Li}(\text{Cu}_{1-x}\text{Ag}_x)_2\text{O}_2$

LCR-

-

4090

Motech

GOM-802

(AC)

(DC)

.

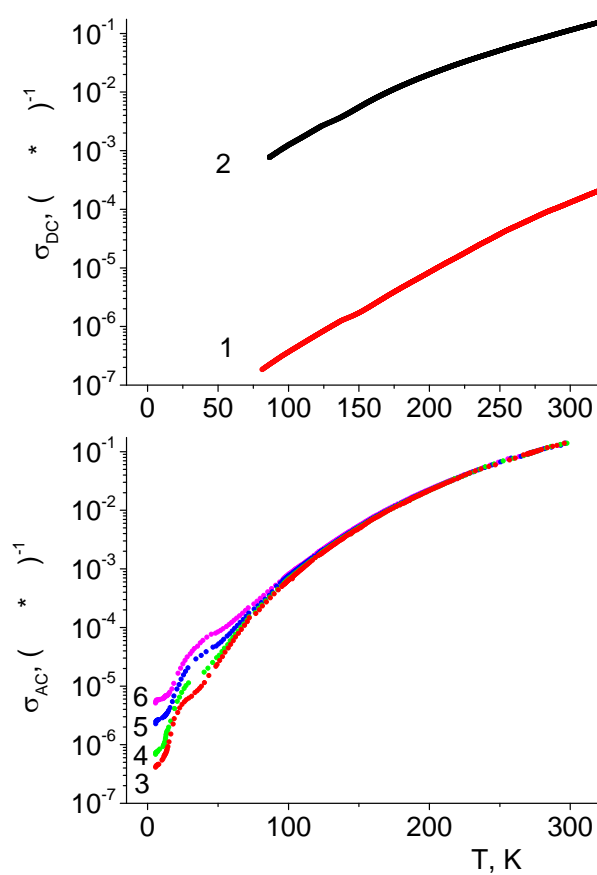
-

[80].

,

 $\text{Li}(\text{Cu}_{1-x}\text{Ag}_x)_2\text{O}_2$ Ag $x = 0,15$ ~ 3

.



. 3.36

c $x=0$ (1) 0,15 (2 - 6),

(1, 2)

10 (

3), 20

(4), 50

(5) 100

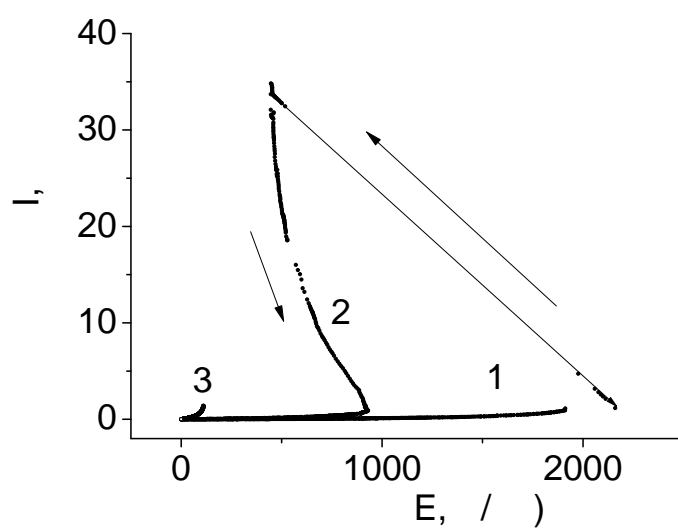
(6)

c

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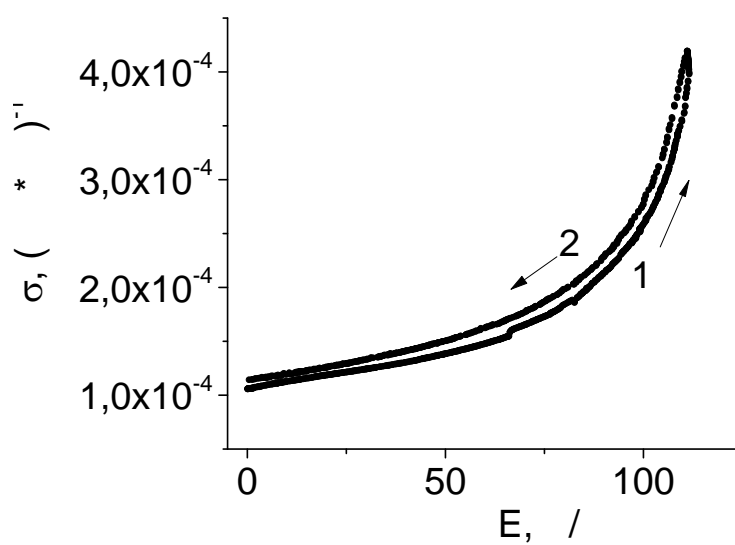
 $T < 240 \text{ K}$ $T \sim 25 \text{ K}$ $\sigma_{AC}(T)$

(3.36).

Ag ($x = 0,05$),

3.37.

Li(Cu_{1-x}Ag_x)₂O₂ = 0,05
 (1, 2) 0,15 (3), (1, 2)
 (3) c 78 .



3.38.

Li(Cu_{1-x}Ag_x)₂O₂ = 0,15

,

 ab 78 .

S-

(. 3.37),

LiCu_2O_2 [29].

$x > 0,05$

, S

—

,

,

:

4

100 / (. 3.38).

3.5.3.2

$\text{Li}(\text{Cu}_{1-x}\text{Zn}_x)_2\text{O}_2$

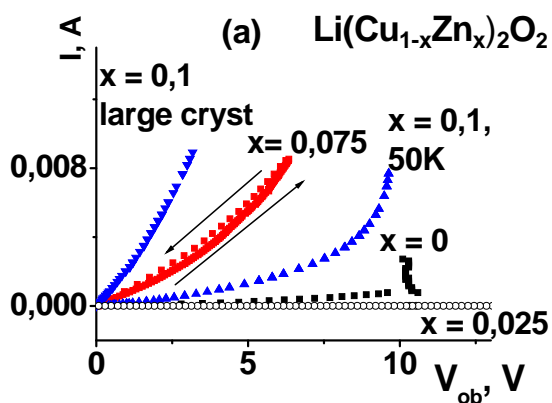
$\text{Li}(\text{Cu}_{1-x}\text{Zn}_x)_2\text{O}_2$

(. 3.39).

Zn

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(. 3.39).



3.39. ()

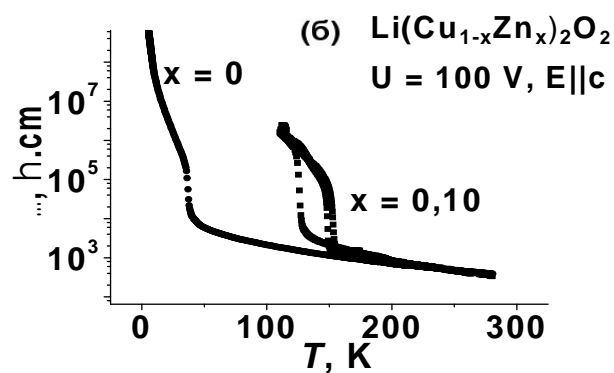
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1

$\text{Li}(\text{Cu}_{1-x}\text{Zn}_x)_2\text{O}_2$

(T)

$\text{Li}(\text{Cu}_{1-x}\text{Zn}_x)_2\text{O}_2$,



100 .

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1. LiCu_2O_2
,
 $\text{Li}_2\text{CuO}_2\text{-CuO}_x$
.
2. - LiCu_2O_2
 $(\text{Li,Ag})\text{Cu}_2\text{O}_2$, $\text{Li}(\text{Cu,Zn})_2\text{O}_2$,
 $3 \cdot 10^{-10}$.
3. , ,
LCO 12 .% Zn (
Cu) 4% Ag (Li).
.
4. $M(T)$,
.
,
O LiCu_2O_2 ($H = 10$)
=150 $\mathbf{M} \parallel$.
O .
Ag $(\text{Li}_{1-x}\text{Ag}_x)\text{Cu}_2\text{O}_2$
($T < 50$) ,
. $x = 0,05$
150 ,
.
5. — - LCO
4,2 - 300 0,1 – 10,0 .
5 . , ~300

$$(\sigma = \sigma_0 \exp(-E_a/kT), \quad E_a = 0,35 - 0,44 \text{ eV})$$

$$(\sigma = \sigma_0 \exp(-T_0/T^{1/4}), \quad T_0 = 10^6 - 10^8 \text{ K}), \quad 25$$

$$E = 5 - 6 \text{ eV}$$

.

$$\lg \sigma \sim 1/T$$

$$\lg \sigma \sim T^{1/4}.$$

5 .

LCO,

,

$$\sigma_a : \sigma_b : \sigma_c = 2 : 1 : 10^4. (\sigma_0 = 295 \text{ s}^{-1}).$$

5 .

-

$$(T, f) \quad \text{tg} \quad (T, f)$$

,

,

$$- E_a = 60 - 79 \text{ K},$$

$$f_r = 10^6 \text{ Hz}, \quad E_a = 1300 \text{ K}, \quad f_r = 2 \cdot 10^8 \text{ Hz}$$

$$(\sigma_0 = 10^4 \text{ s}^{-1})$$

$$(\sigma_0 = 10^4 \text{ s}^{-1}),$$

.

6.

Ag Zn

LiCu₂O₂

:

$$x > 0,05$$

$$\sim 3$$

.

$$x < 0,05$$

;

$$c \quad x > 0.05$$

,

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7.

LCO

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LCO

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