

LiCu₂O₂,

01.04.07

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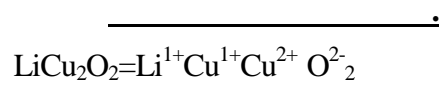
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LCO

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LCO

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LCO,

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LiCu_2O_2 ,

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LiCu_2O_2 ,

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LiCu_2O_2

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- LiCu₂O₂
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- :
1. Li₂CuO₂–CuO_x;
 - LiCu₂O₂
 Li(Cu,Zn)₂O₂, (Li,Ag)Cu₂O₂;
 4 10 10 ; LiCu₂O₂.
 2. Li(Cu_{1-x}Zn_x)₂O₂, (Li_{1-x}Ag_x)Cu₂O₂
 $= 0 - 0,12 \quad = 0 - 0,04,$ Zn Ag
 LiCu₂O₂,
 Li(Cu_{1-x}Zn_x)₂O₂, (Li_{1-x}Ag_x)Cu₂O₂.
 3. $\sigma_{DC},$
 $() = \sigma_{AC}$
 LiCu₂O₂ Li(Cu,Zn)₂O₂, (Li,Ag)Cu₂O₂ 4,2 – 300 0,1 – 100 .
 3 . , LiCu₂O₂ Li(Cu,Zn)₂O₂,
 (Li,Ag)Cu₂O₂ $\sigma_{DC} \quad T \sim 300 \text{ K}$
 - $(\sigma_{DC} = \sigma_0 \exp(E_a/k_B T))$
 $() ,$ $\sigma_{DC} = A \exp(T_0/T)^{1/4}.$
 ~25 DC

3 . LiCu_2O_2

$a, b, c: \dagger_a : \dagger_b : \dagger_c = 2 : 1 : 10^4$.

3 .

$\text{Li}(\text{Cu}_{1-x}\text{Zn}_x)_2\text{O}_2$, $(\text{Li}_{1-x}\text{Ag}_x)\text{Cu}_2\text{O}_2$, $(x(\text{Zn}) = 0,05$,
 $x(\text{Ag}) < 0,02)$

S- - .

4. LCO

O_δ

LCO, O_δ ,

1 – 2 ,

150 K. O_δ

_____.

LiCu_2O_2

$\text{Li}(\text{Cu,Zn})_2\text{O}_2$, $(\text{Li,Ag})\text{Cu}_2\text{O}_2$:

LiCu_2O_2

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 ,
 CuO₄-
 (ladder compound),
 ,

LiCu₂O₂

LiCu₂O₂

: 1) -Cu¹⁺(1)-, 2) -O(1)Cu²⁺(2)O(2)Li- 3) -

LiO(2)Cu²⁺(2)O(1) [1] (. 1). Cu¹⁺ ,

O²⁻-Cu¹⁺-O²⁻ , LiCuO₂- . 2) 3)

CuO₅ LiO₅, *ab*-

LiO₄ CuO₄ , *a* ,

Cu-O- Li-O- , *b* -

Cu-O- Li-O- . 1) Cu¹⁺ .

LiCu₂O₂ -

Cu²⁺-O , -LiCuO₂- ,

, *b*.

-

Li¹⁺ Cu¹⁺ ,

.

Cu-O- ,

[1]). LiCu₂O₂

.

$T_{c1} = 24,6$ T_{c2}

$= 23,2$:

T_{c1} T_{c2} T_{c2} [1].

c-

P_s . T_{c2} ,

[1].

LiCu₂O₂ ,

1) Cu¹⁺ ,

- O²⁻ , p- (. 1).

O²⁻

Cⁿ⁺ = Cu²⁺ Li¹⁺ ,

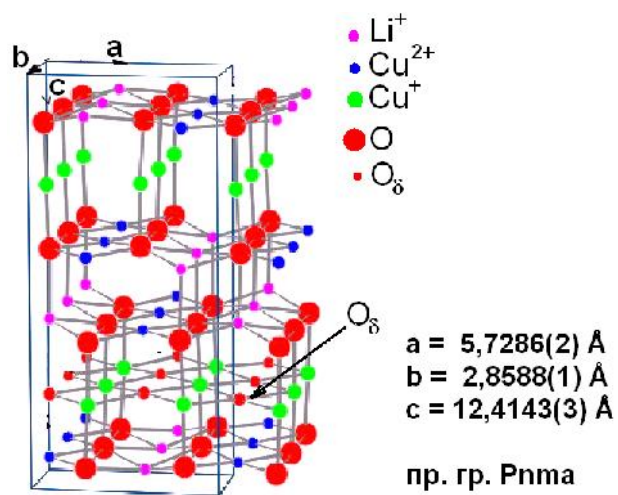
CO₅ CO₆.

O₂p-, Cu_{3d}-O₂p- CuO₄.

Pmna

[3]. ,

LCO - Cu²⁺



1. LiCu_2O_2 ([1]).

Li^+ $\text{u}(\text{Li})\text{-O-Cu}(\text{Li})$, - LCO.

LiCu_2O_2 ,

LiCu_2O_2 $(\text{Li},\text{Na})\text{Cu}_2\text{O}_2$ [4],

$\text{Li}(\text{Cu},\text{Zn})_2\text{O}_2$ $\text{Li}(\text{Cu},\text{Ni})_2\text{O}_2$ [5].

LiCu_2O_2

,

: 1) -

2)

-2- ().

()

-4

: $(\text{CuK}_1) = 1,54051$, $(\text{CuK}_2) = 1,54433$, $(\text{CuK}_{\text{av}}) = 1,54178 \text{ \AA}$.

“Orbis” “EDAX” () Si(Li)
 , Na.
 () Q1500
 D F. Paulik, J. Paulic, L. Erdey.

7-20 100 – 100 ,
 ,
 $R_1 = 5,26$,
 5 – 300 SQUID
 MPMS-XL-7 Quantum Design Inc ($H = 20$)
 (zero- field-cooled – ZFC) (field-cooled – FC).

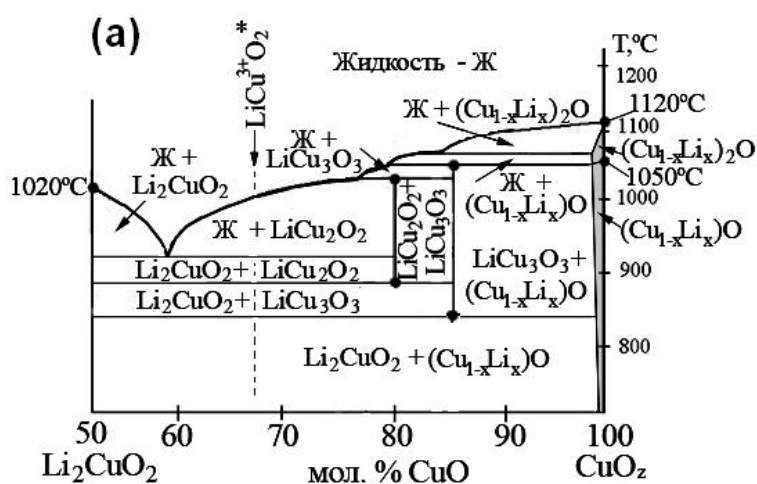
LiCu₂O₂ ,
 .

3.1. , *Li₂CuO₂ - CuO* .

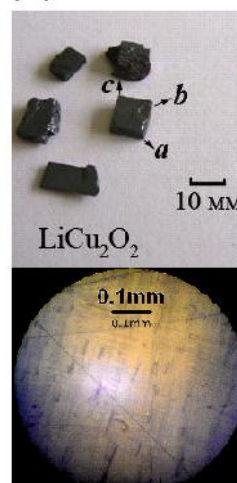
(. 2).
 : LiCu₂O₂ LiCu₃O₃,
 (1323 1373) (1163 1113
).

1173 - 1323

LiCu₂O₂.
 CuO « . ». Li₂CO₃ « . ».
 LiCu₂O₂ xCuO·(1-x)Li₂CO₃ c 0,77 x 0,83 4 1393 0,5
 , 1323 ,
 2,0 / - 1173 ,
 1173 10 - 20



(б)



2.) $\text{Li}_2\text{CuO}_2 - \text{CuO}_z$
 ;) LiCu_2O_2 ()
 LiCu_2O_2 , () (001).

1393

1323

~1173

 LiCu_2O_2 .

LiCu_2O_2 , (0,5 – 4) $8 \times 8 \times 3$ (. 2).
 (001),

(210) (2-10),

{001}

{210}.

 LiCu_2O_2 ,

(120) (1-20) (. 2).

 LiCu_2O_2 LiCu_2O_2

Ag

Zn

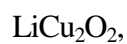
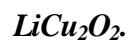
LCO.

$\text{Li}_2\text{CO}_3 \cdot 4(1-x)\text{CuO} \cdot 4x\text{AgNO}_3$, $\text{Li}_2\text{CO}_3 \cdot 4(1-x)\text{CuO} \cdot 4x\text{ZnO}$ - I (1-x) $\text{Li}_2\text{CO}_3 \cdot 2x\text{ZnO} \cdot 4\text{CuO}$ - II 0 x 0,5,

 Li_2CO_3 , CuO, AgNO₃, ZnO

« », « », « » « »

, LCO,
,
,
> 0,15 1 .



6

90 .

5,0 / .

6 20 . *a*

,



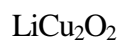
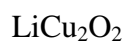
~1113

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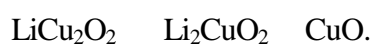
400

1 - 4 .

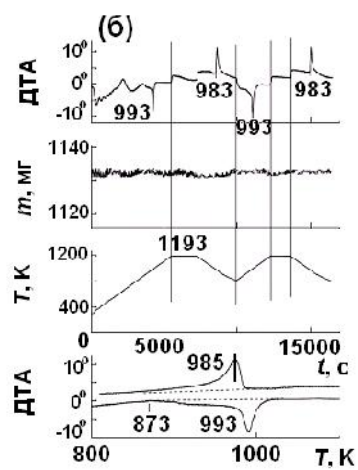
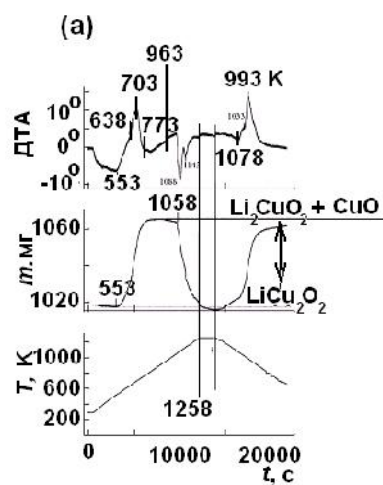


[3],

$T = 553-773$



m



3.

$LiCu_2O_2$:)

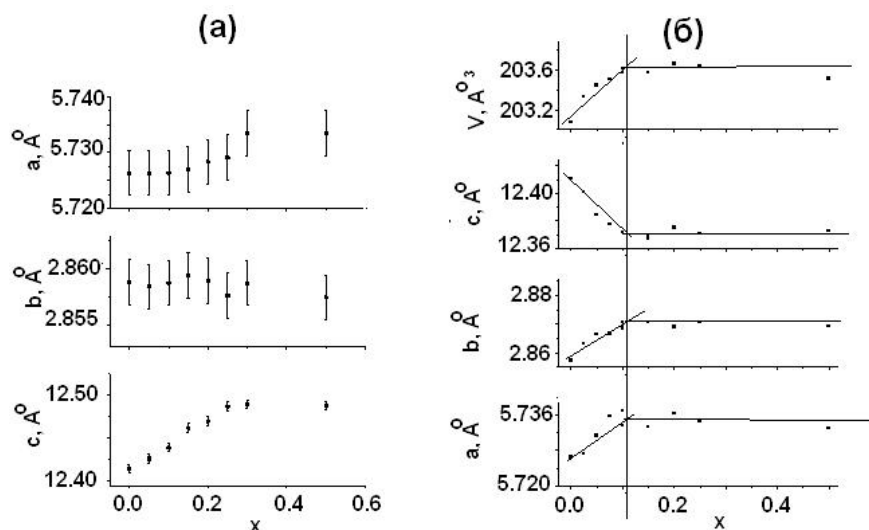
)

(. 3) Cu^+ Cu^{2+} ,
 $\text{LiCu}^{2+}\text{Cu}^+\text{O}_2$.
1073–1223 LiCu_3O_3 ,
 LiCu_2O_2 , .
 $\text{Li}(\text{Cu}_{0,95}\text{Zn}_{0,05})_2\text{O}_2$ LiCu_2O_2 .

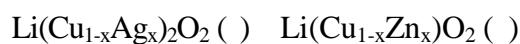
LiCu_2O_2			
1	-		2 400
2	-		-
1			-
2			-
A4	-	40 . 1113	2 4 400
W2	-	40 . 1113	

LiCu_2O_2 LiCu_2O_2 .
 LiCu_2O_2 (1320). $T_{\text{PT}} = 993$
, (. 3).
,
.
 $T = 983$
,
.
 LiCu_2O_2 .

: $a = 5,73$, $b = 2,86$, $c = 12,41 \text{ \AA}$,
[1] LiCu_2O_2 .
 ZnO , Zn
 LiCu_2O_2 ((Li,Zn) Cu_2O_2 - I, $\text{Li}(\text{Cu,Zn})_2\text{O}_2$ – II II.
. 4 a, b, c
 $\text{Li}_2\text{CO}_3 \cdot 4(1-x)\text{CuO} \cdot 4x\text{AgNO}_3$
 $\text{Li}_2\text{CO}_3 \cdot 4(1-x)\text{CuO} \cdot 4x\text{ZnO}$. Ag



4.

 a c b $x = 0,25$. $x = 0,25$

(. 4).

Zn

 a, b c $= 0 - 0,12, > 0,12$

(. 4).

-

().

-

 $[\text{Cu}]/[\text{Li}]$

,

2

-

,

 $[\text{O}]/[\text{Li}]$ $2,2 - 2,3 = 2 +$

,

 $= 0,2 - 0,3$

O ,

,

123,

[5].

,

Ag

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4 .%

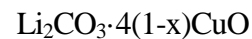
Ag

 $x = 0,25$.

.

Zn

,

 $4x\text{ZnO} = 0,10$

12 .% (

).

,

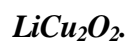
,

 $\text{Li}_2\text{u}_2\text{O}_2$

4 .% Ag 12 .% Zn.



3.2.



$$\langle \sigma \rangle = M/H, \quad (H = 10^{-10}),$$

40 ,

- (. 5).

,

2

—

•

3

 Cu^{1+} ,

1,

$$\text{Cu}^+$$

Cu^{2+}

$$S = 1/2.$$

$(T, H).$

O,

•

$$(T) \quad \begin{array}{lll} \text{DC} = 10 & 1 \text{ (} \cdot 5 \text{)}. & < 400 \text{ K,} \\ \text{O, (} & & \text{ ()}, \end{array}$$

• ,

5 (T), (H = 10). ,

.

(T)	148	124,7	2	2,	,
		(T)	H	b	2

2

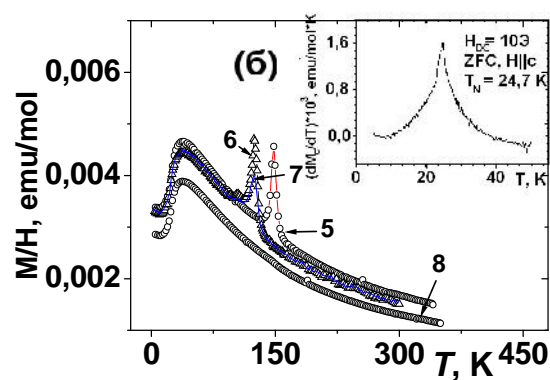
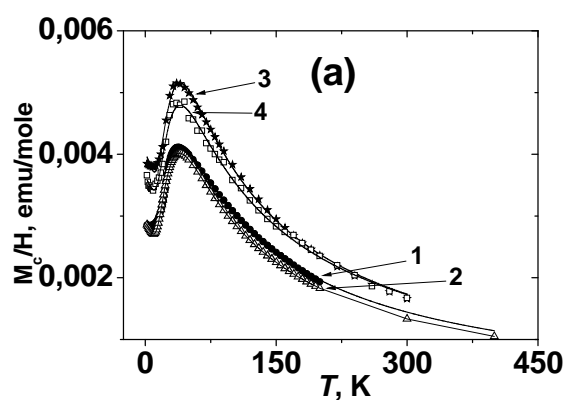
•

(FC, ZFC).

O

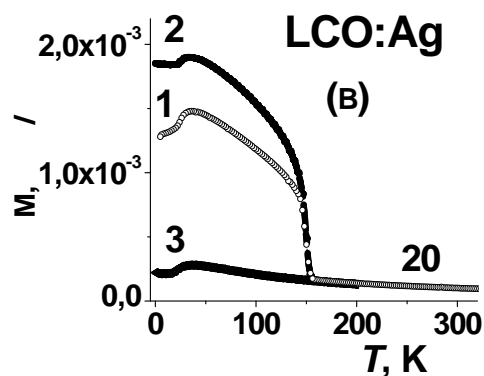
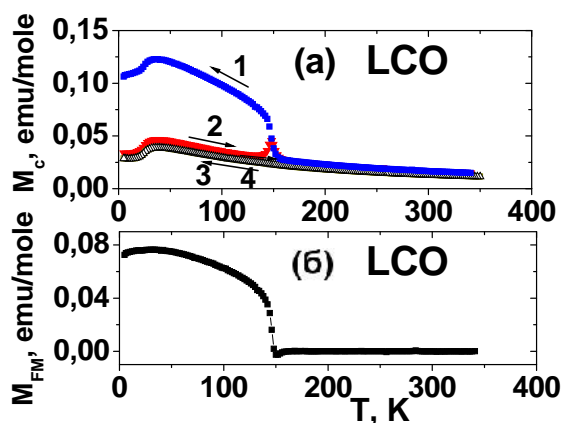
$$(H = 10) = 150 \text{ LiCu}_2\text{O}_{2+}$$

$$(= 5)$$



5. () \parallel $H_{DC} = 10$, () :
 (1 2) $f = 110$ $h_{ac} = 2$ (3, 4) ;
 () $H_{DC} = 10$, $6 -$ 2 ZFC: 5, 8 -
 918 , 1 $H_{DC} = 0,5$, $7 -$, $H_{DC} = 10$ (5, 6, 7
 $H \parallel c$, 8 - $H \parallel b$ $H_{DC} = 10$).
 $dM_c(T)/dT$ 2, $N = 24,7$
 $LiCu_2O_2$. **c.**

·
 O
 ,
 [6] -
 $LiCu_2O_2$: CuO_6
 ,
 ,
 [210].
 ,
 CuO_4 [7],
 - [8]
 +1.
 , CuO_4
 ·
 ·
 · 6() , **c**
 b a (, a
 b).



6. M_{FM} , emu/mole M_c , emu/mole

(1) FC; (2) ZFC; (3) ZFC; (4) FC.)

10 : (3) ZFC; (4) FC.)

$M_{FM} = M_{FC} - M_{ZFC}$ ()

2. $Li(Cu_{1-x}Ag_x)_2O_2$ = 0 (1), 0,05 (2) 0,15 (3)

() , 20 .

$LiCu_2O_2$

$Li_2O \cdot 4(1-x)CuO \cdot 4AgNO_3$ c 0 x 0,5, $M(T)$

5–300 (H=20), c

ZFC FC. ZFC FC $M(T)$

$T = 37$ (. 6),

T_{c1} T_{c2}

$dM(T)/dT$.

$M(T)$

$LiCu_2O_2$.

F

$M(T)$

$x = 0,05$,

,

$T_3 = 150$.

T_3

ZFC FC.

Ag ($x > 0,05$)

$T_{c3} = 150$

(. 6).

3.3.

$LiCu_2O_2$.

$\sigma(\omega=0) = \sigma_{DC}$

LCO (. 8)

$T > 300$ K

$\sigma_{DC} = \sigma_o \exp(-E_a/kT)$

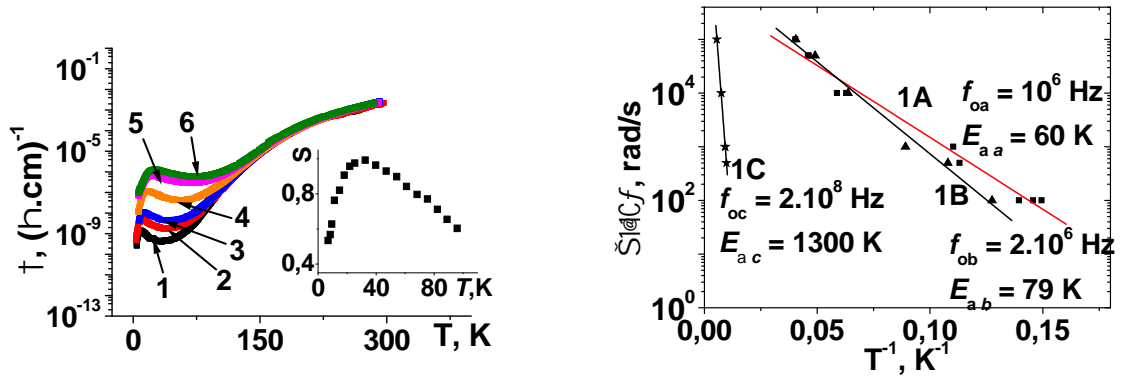
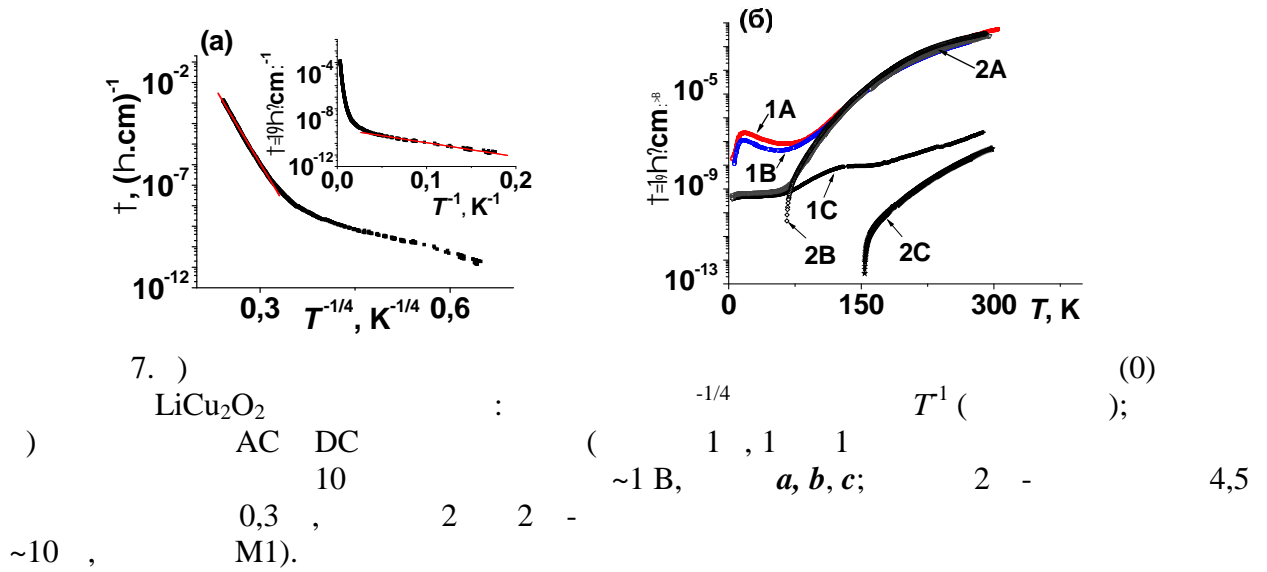
E_a , 0,35 - 0,44 (), 100 - 300 -

$\sigma_{DC} = \sigma_o \exp(-T_o/T)^{1/4}$ $T_o = 10^6 - 10^8$ K ().

25

DC

:



8.) (T,) LiCu_2O_2 b (1, 2, 3, 4, 5 6) s () ().

9. M1 $a - 1A, b - 1B, c - 1C$.

$a \sim 5 - 6$,

(. 8) -

$T \sim 295$ c , ab ,

b , a .

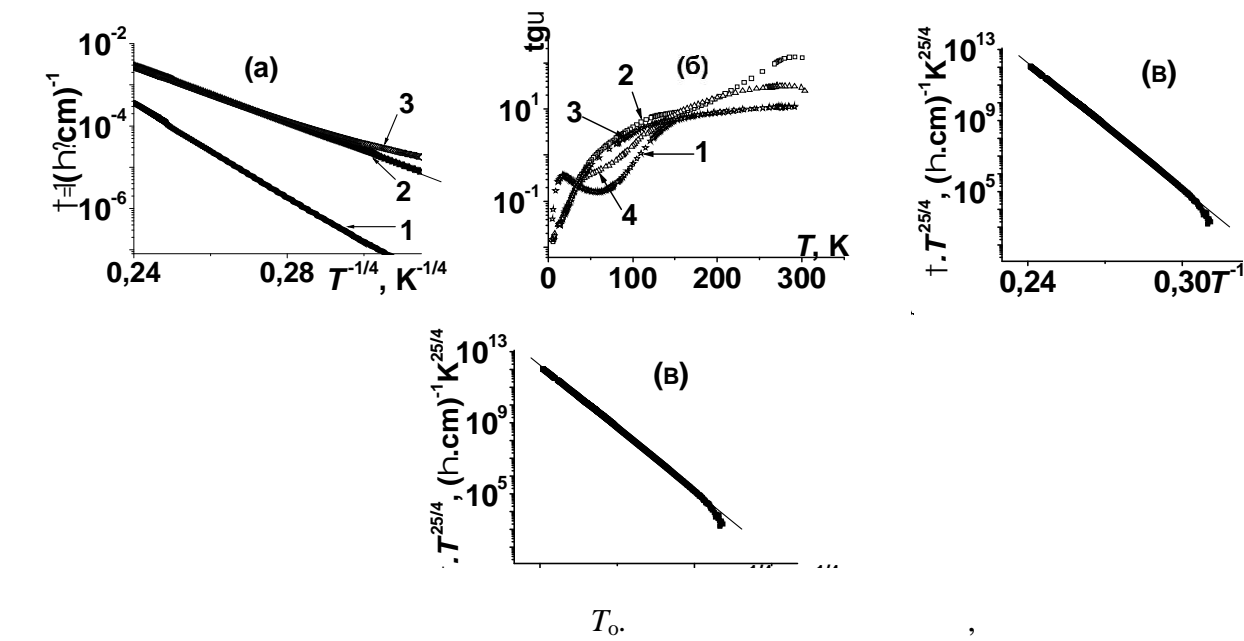
$E_D = -d(\ln)/d(1/T)$,

~ 200 E_D 0,15; 0,12 0,1

0,3; 107 150 ,

($\ln \sim 1/T$) ($\ln \sim T^{1/4}$).

$T < 30$ ***ab***,
 , ***a*** ***b***
16,6 15,7 , ***c***
- ~134 .
 .
. 8 (T ,), M1 ***b***
 , ~100
80 , 80
 , (T ,) = T^n s ~1,
()
 .
 $s(T)$ () = 30 ,
 .
 .
 , $\text{tg } (T)$
a, ***b***, ***c***, , $\lg f - 1/T_{\max}$,
(. 9).
($= 1/2 f$), =
 $\exp(E_{\text{ai}}/kT)$, $i = \textbf{\textit{a}}, \textbf{\textit{b}}, \textbf{\textit{c}}$. f_{oi} ,
 ,
 $\sim 10^6$ ***a*** ***b***, $2 \cdot 10^8 -$. $f_{\text{oa,b}}$
(), $E_{\text{a} \textbf{\textit{a}}, \textbf{\textit{b}}}$
 CuO_2 (***b***) () [1].
c f_o
(). , $< 30 \text{ K}$
 , $> 30 \text{ K} -$.
 .
DC 295 ***ab*** 20%, ***c*** -
 .
 - (),
 , ***c*** , , ,
 . T_o
(. 10).



tg (T) A4

20 , ; 200

(. 10).

>> 1

DC

ab

$\sim 10^{-10}$ (.)⁻¹ ~50 , - ~100 . ,

(). , ab

m = 25/4 T₀ = 6,75.10⁷ (. 10),

[9], , , Li⁺ Cu²⁺

LiCu₂O₂₊ .

LiCu₂O₂



~3

DC

$$T < 240 \text{ K}$$

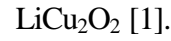
$$T \sim 25 \text{ K}$$

$$\sigma_{AC}(T)$$

$$\text{Ag} (x = 0,05),$$

S-

(. 11a),

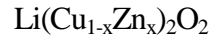


$$x > 0,05$$

, S

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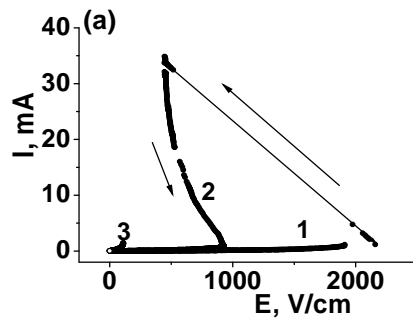
$$100 / (. 11).$$



$$(. 12).$$

Zn

$$(. 12).$$



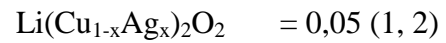
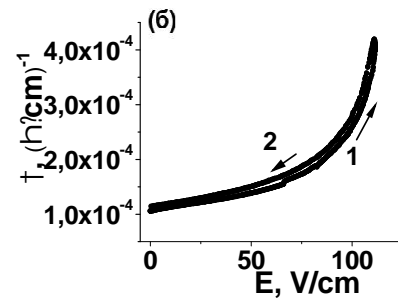
11.)

0,15 (3),

(1, 2)

).

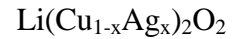
=0,15 (



(3)

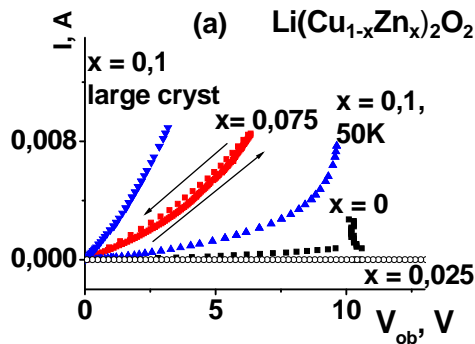
c

78 .

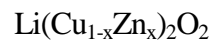


ab

78).

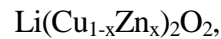


12.)



)

(T)



;

1

100 .

1. LiCu_2O_2
2. $\text{Li}_2\text{CuO}_2\text{-CuO}_x$
3. LiCu_2O_2 $(\text{Li,Ag})\text{Cu}_2\text{O}_2$, $\text{Li}(\text{Cu,Zn})_2\text{O}_2$
4. $M(T)$, LCO
5. LiCu_2O_2 $(H = 10 \text{ Oe})$ $= 150$ $\text{M} \parallel$
6. $\text{Ag} (\text{Li}_{1-x}\text{Ag}_x)\text{Cu}_2\text{O}_2$ $(T < 50 \text{ K})$
7. $x = 0,05$
8. 150 K
9. LCO $4,2 - 300$ $0,1 - 10,0$
10. ~ 300
11. $(\sigma = \exp(-E_a/kT), E_a = 0,35 - 0,44 \text{ eV})$

Roessli, U. Staub, A. Amato [] // Physica B. 2001. V. 296. – P.306–311.

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 , , // 2008. . 44. –
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 / // . 1999. . 21, . 2. – .
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1. DC AC
 LiCu₂O₂₊ / , , [] // .
 « ». 2013. – 2. – . 174–178.

2. ,
 LiCu₂O₂ / ,
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